

# Sediment pore water and solid phase geochemical analyses from samples collected on R/V Savannah cruise SAV-16-16 at the Cape Lookout Continental Shelf and Slope during May 2016

**Website:** <https://www.bco-dmo.org/dataset/806065>

**Data Type:** Cruise Results

**Version:** 1

**Version Date:** 2020-03-17

## Project

» [Source, Composition, and Stability of Soluble Iron Fluxing from Continental Margin Sediments](#) (Soluble Fe Fluxing)

Contributors	Affiliation	Role
<a href="#">Taillefert, Martial</a>	Georgia Institute of Technology (GA Tech)	Principal Investigator
<a href="#">Rauch, Shannon</a>	Woods Hole Oceanographic Institution (WHOI BCO-DMO)	BCO-DMO Data Manager

## Abstract

Sediment pore water and solid phase geochemical analyses from samples collected on R/V Savannah cruise SAV-16-16 at the Cape Lookout Continental Shelf and Slope during May 2016.

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## Coverage

**Spatial Extent:** N:34.635 E:-75.799 S:33.869 W:-76.605

**Temporal Extent:** 2016-05-17 - 2016-05-26

## Methods & Sampling

Sediment cores were obtained by a MC-800 multi-corer and profiled immediately with voltammetric Hg/Au microelectrodes deployed on a computer-controlled micromanipulator (Beckler et al., 2016). Sediment pore waters were then extracted from the same core and either preserved until analysis or analyzed immediately onboard ship. Sampling and analyses were conducted immediately after sediment collection to minimize artifacts from exposure to the atmosphere.

After profiling, sediments were immediately sliced under N<sub>2</sub> atmosphere and pore waters extracted by centrifugation at 3000 rpm under N<sub>2</sub> atmosphere. Finally, pore waters were immediately filtered (0.2 µM PSE Puradisc syringe filters, Whatman) under N<sub>2</sub> atmosphere and either preserved at -20C until analysis (SO<sub>4</sub><sup>2-</sup>), dispensed directly into reagents for analysis (SPO<sub>4</sub><sup>3-</sup>), or analyzed immediately (DIC). SO<sub>4</sub><sup>2-</sup> was measured by non-suppressed HPLC with UV detection (Beckler et al., 2014). SPO<sub>4</sub><sup>3-</sup> was measured spectrophotometrically using the molybdate-blue method after natural color correction to avoid interferences from dissolved silica and sulfides (Murphy and Riley, 1962). Finally, DIC was measured by flow injection analysis with conductivity detection after spiking samples with 10 mM ZnCl<sub>2</sub> to prevent dissolved sulfide interferences (Hall and Aller,

1992). All errors reported for the electrochemical measurements represent the standard deviation of at least triplicate measurements. Errors of all other parameters represent the analytical error propagated from calibration curves, dilution, and instrumental drift

**Problem report:** Some of the data are missing because pore water volumes were too low to make all the measurements.

## Data Processing Description

Chromatographic data were processed using Voltint (Bristow and Taillefert, 2008), a Matlab<sup>TM</sup>-based software developed for these applications. Spectrophotometric measurements were processed manually.

### BCO-DMO Processing:

- added date column using original year, month, and day columns.

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## Data Files

File
<b>porewater_SAV16-16.csv</b> (Comma Separated Values (.csv), 16.45 KB) MD5:cb1cb2ddd0cfcb2b74d659359ca4d970
Primary data file for dataset ID 806065

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## Related Publications

Beckler, J. S., Kiriazis, N., Rabouille, C., Stewart, F. J., & Taillefert, M. (2016). Importance of microbial iron reduction in deep sediments of river-dominated continental-margins. *Marine Chemistry*, 178, 22–34.

doi:[10.1016/j.marchem.2015.12.003](https://doi.org/10.1016/j.marchem.2015.12.003)

*Methods*

Beckler, J. S., Nuzzio, D. B., & Taillefert, M. (2014). Development of single-step liquid chromatography methods with ultraviolet detection for the measurement of inorganic anions in marine waters. *Limnology and Oceanography: Methods*, 12(8), 563–576. doi:[10.4319/lom.2014.12.563](https://doi.org/10.4319/lom.2014.12.563)

*Methods*

Bristow, G., & Taillefert, M. (2008). VOLTINT: A Matlab<sup>®</sup>-based program for semi-automated processing of geochemical data acquired by voltammetry. *Computers & Geosciences*, 34(2), 153–162.

doi:[10.1016/j.cageo.2007.01.005](https://doi.org/10.1016/j.cageo.2007.01.005)

*Methods*

Eitel, E. M., Owings, S. M., Belli, K. M., Beckler, J. S., Williams, A., Fields, B. P., ... Taillefert, M. (2020). Variations in sediment production of dissolved iron across a continental margin not dominated by major upwelling or riverine inputs. *Marine Chemistry*, 220, 103750. doi:[10.1016/j.marchem.2020.103750](https://doi.org/10.1016/j.marchem.2020.103750)

*Results*

Hall, P. . J., & Aller, R. C. (1992). Rapid, small-volume, flow injection analysis for total CO<sub>2</sub>, and NH<sub>4</sub><sup>+</sup> in marine and freshwaters. *Limnology and Oceanography*, 37(5), 1113–1119. doi:[10.4319/lo.1992.37.5.1113](https://doi.org/10.4319/lo.1992.37.5.1113)

*Methods*

Murphy, J., & Riley, J. P. (1962). A modified single solution method for the determination of phosphate in natural waters. *Analytica Chimica Acta*, 27, 31–36. doi:[10.1016/s0003-2670\(00\)88444-5](https://doi.org/10.1016/s0003-2670(00)88444-5)

*Methods*

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## Parameters

Parameter	Description	Units
Year	Year	unitless
Month	Month	unitless
Day	Day	unitless
Collection_Type	Instrument used for collection	unitless
Station	Station number	unitless
Lon	Longitude	decimal degrees East
Lat	Latitude	decimal degrees North
sample_ID	Sample ID number	unitless
Sediment_depth	Sediment depth	centimeters (cm)
SO4	Sulfate	millimolar (mM)
sdSO4	Standard deviation of SO4	millimolar (mM)
PO4	Dissolved orthophosphate	micromolar (uM)
sdPO4	Standard deviation of PO4	micromolar (uM)
DIC	Dissolved inorganic carbon	millimolar (mM)
sdDIC	Standard deviation of DIC	millimolar (mM)
date	Date; format: yyyy-mm-dd	unitless

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## Instruments

<b>Dataset-specific Instrument Name</b>	Flow Injection Analysis
<b>Generic Instrument Name</b>	Flow Injection Analyzer
<b>Dataset-specific Description</b>	Flow Injection Analysis with peristaltic pump (Gilson), conductivity detector (Fisher Scientific), and integrator with LC-100 software (Analytical Systems, Inc.)
<b>Generic Instrument Description</b>	An instrument that performs flow injection analysis. Flow injection analysis (FIA) is an approach to chemical analysis that is accomplished by injecting a plug of sample into a flowing carrier stream. FIA is an automated method in which a sample is injected into a continuous flow of a carrier solution that mixes with other continuously flowing solutions before reaching a detector. Precision is dramatically increased when FIA is used instead of manual injections and as a result very specific FIA systems have been developed for a wide array of analytical techniques.

<b>Dataset-specific Instrument Name</b>	HPLC
<b>Generic Instrument Name</b>	High-Performance Liquid Chromatograph
<b>Dataset-specific Description</b>	SO <sub>4</sub> <sup>2-</sup> was measured by non-suppressed HPLC with UV detection
<b>Generic Instrument Description</b>	A High-performance liquid chromatograph (HPLC) is a type of liquid chromatography used to separate compounds that are dissolved in solution. HPLC instruments consist of a reservoir of the mobile phase, a pump, an injector, a separation column, and a detector. Compounds are separated by high pressure pumping of the sample mixture onto a column packed with microspheres coated with the stationary phase. The different components in the mixture pass through the column at different rates due to differences in their partitioning behavior between the mobile liquid phase and the stationary phase.

<b>Dataset-specific Instrument Name</b>	MC-800 multi-corer
<b>Generic Instrument Name</b>	Multi Corer
<b>Dataset-specific Description</b>	Sediment cores were obtained by a MC-800 multi-corer.
<b>Generic Instrument Description</b>	The Multi Corer is a benthic coring device used to collect multiple, simultaneous, undisturbed sediment/water samples from the seafloor. Multiple coring tubes with varying sampling capacity depending on tube dimensions are mounted in a frame designed to sample the deep ocean seafloor. For more information, see Barnett et al. (1984) in Oceanologica Acta, 7, pp. 399-408.

<b>Dataset-specific Instrument Name</b>	Spectrophotometer Thermofisher
<b>Generic Instrument Name</b>	Spectrophotometer
<b>Dataset-specific Description</b>	PO <sub>4</sub> <sup>3-</sup> was measured spectrophotometrically
<b>Generic Instrument Description</b>	An instrument used to measure the relative absorption of electromagnetic radiation of different wavelengths in the near infra-red, visible and ultraviolet wavebands by samples.

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## Deployments

### SAV-16-16

<b>Website</b>	<a href="https://www.bco-dmo.org/deployment/805919">https://www.bco-dmo.org/deployment/805919</a>
<b>Platform</b>	R/V Savannah
<b>Start Date</b>	2016-05-15
<b>End Date</b>	2016-05-30
<b>Description</b>	More information from the Rolling Deck to Repository (R2R): <a href="https://www.rvdata.us/search/cruise/SAV-16-16">https://www.rvdata.us/search/cruise/SAV-16-16</a>

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## Project Information

### Source, Composition, and Stability of Soluble Iron Fluxing from Continental Margin Sediments (Soluble Fe Fluxing)

**Coverage:** Northern South Atlantic Bight (Cape Lookout, NC) and Northern Gulf of Mexico

#### *NSF Award Abstract:*

Iron is a limiting nutrient in the world's oceans and plays a key role in regulating the growth of phytoplankton. The main sources of iron to the open ocean are the atmosphere, through wind-blown terrestrial dust, and the seafloor, through input from continental shelf sediments. While atmospheric inputs have been well-studied, the oceanic input of iron from sediments has only sparsely been measured and, as a result, the relative importance of the sediment-derived iron to the iron pool and, ultimately, primary productivity in the oceans is poorly understood. In this study, researchers will examine the chemical properties of sediment-derived iron in the oceans to assess its contribution to the iron used by phytoplankton. Results from this study will further our understanding of iron inputs to the ocean and their importance to ocean primary productivity. The project will contribute to the training of graduate students, as well as provide educational opportunities such as a day at sea for undergraduate students in engineering and physical science.

The atmosphere and continental margin sediments are the main source of the limiting nutrient iron (Fe) to the open ocean. Yet, the chemical form of iron from sediments has not been well examined and only quantified as reduced iron or the dissolved iron passing through 0.45 µm filters. The kinetics of iron oxygenation suggests it should precipitate rapidly in the overlying waters, challenging the view that sediments are important sources of iron for primary production. To establish whether the flux of iron from sediments has important implications for primary productivity, possibly rivaling atmospheric inputs, it is necessary to demonstrate that ferric iron originating in sediments is under the form of stable iron species with potential for a high residence time in the

water column. The overall objective of this project is to test the hypotheses that iron fluxing across the sediment-water interface in continental margin sediments is dissolved under the form of organic-Fe(III) complexes and that the magnitude of the iron flux is influenced by the redox conditions in the overlying waters, the composition of the complexes, and the biogeochemical processes in the underlying sediments. To test these hypotheses, the flux and speciation of dissolved Fe(III) will be quantified in the sediments of the Carolina depocenter and the Gulf of Mexico, and the biogeochemical processes regulating the production and the flux of iron as a function of the redox regime of the environment will be determined using in situ measurements and state-of-the-art voltammetric and chromatographic techniques.

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## Funding

Funding Source	Award
<a href="#">NSF Division of Ocean Sciences (NSF OCE)</a>	<a href="#">OCE-1438648</a>

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