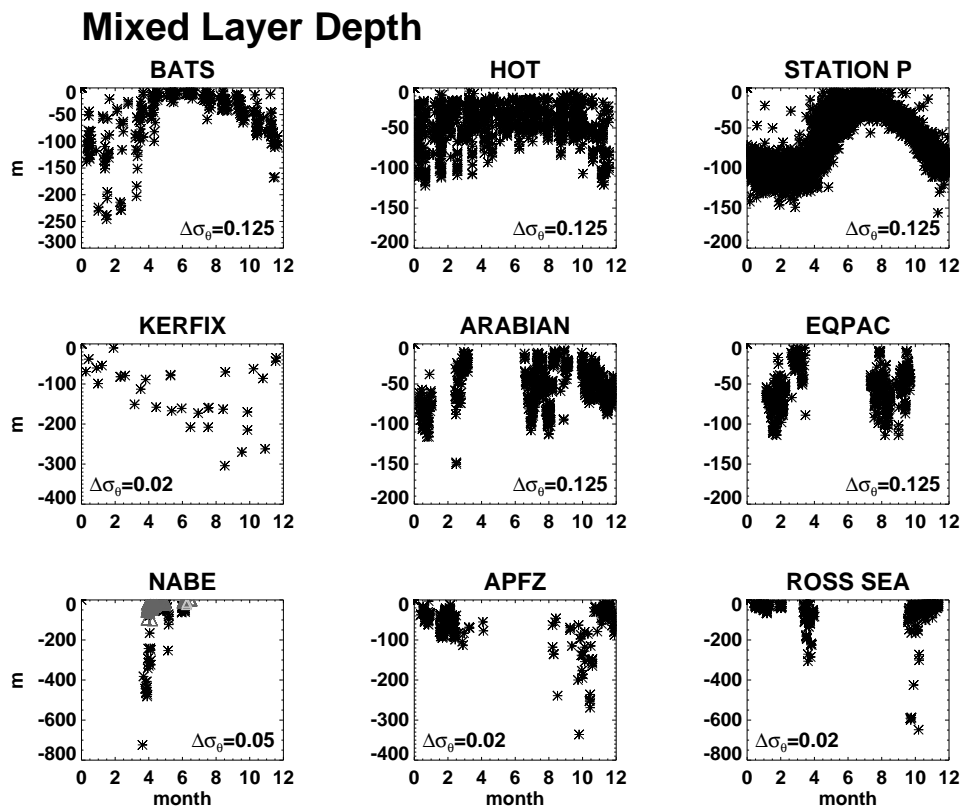


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# Nutrients, Chlorophyll, Primary Production and Related Biogeochemical Properties in the Ocean Mixed Layer

A Compilation of Data Collected at Nine JGOFS Sites

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## Preface

Field measurements of nutrients, chlorophyll, primary production and other biogeochemical parameters are compiled for nine JGOFS time-series and process study sites. Daily mixed layer depths are determined for each site and then used to calculate average mixed layer concentrations for each available parameter. The data span a range of biogeochemical regimes from oligotrophic zones, classic spring bloom and high-nitrate low-chlorophyll (HNLC) environments. The resulting data sets should facilitate regional and global ecosystem model development by providing a consistent, quality controlled set of observations for key JGOFS sites. Data are presented for four Time-Series locations: BATS, HOT, KERFIX, and Station P, and four U.S. JGOFS Process Study sites: Arabian Sea, Equatorial Pacific (EqPac), North Atlantic Bloom Experiment (NABE), and the U.S. Southern Ocean Survey (which includes two distinct regions: the Ross Sea and Antarctic Polar Front Zone). These data are described in this report, including a description of data sources, methodologies, and results with a focus on the seasonal cycle composites. The data and documentation are available electronically through the Data Support Section at the National Center for Atmospheric Research (<http://dss.ucar.edu/datasets/ds259.0>).

# 1. INTRODUCTION

Global carbon cycle models are becoming more prevalent and sophisticated as the scientific community seeks to understand how various earth systems will respond to increasing CO<sub>2</sub> in the atmosphere. The role of the biosphere, particularly in the marine environment, is gaining more attention as physical and biological modelers alike realize that the ocean's role in the carbon cycle may be largely influenced by biogeochemical reactions in the upper ocean. The fixing of inorganic carbon by phytoplankton in the euphotic zone (production), largely depends on physical processes to deliver limiting macronutrients from deeper oceanic zones. The ultimate fate of that production is in turn controlled by physics (advection, mixing), and by biological processes such as how various food webs package the carbon into sinking particles or dissolved organic matter (DOM), how much is recycled in the euphotic zone, and how much is exported as sinking material or DOM. Net community export production is a measure of how much fixed carbon is removed from the surface ocean. These values are difficult to obtain on a global or even regional basis, and vary widely from studies based on satellite data (e.g. Field *et al.* 1998), empirical and theoretical temperature relationships (e.g. Laws *et al.* 2000) and ecosystem models (e.g. Moore *et al.* in press(a), in press(b)).

Bringing these various approaches toward a more complete understanding of upper ocean carbon cycle science has been a standing goal of the U.S. JGOFS Synthesis and Modeling Project (U.S. JGOFS 1997). This technical report is an effort to provide investigators with physical, biological and chemical data in a common format for various sites around the globe (Fig. 1). The ideal locations for reporting such data are the time-series stations. These provide regular observations over long time periods and capture both the natural variability and long-term trends at specific locations. In addition, detailed process study investigations have been conducted in most major oceans. These studies were typically confined to a few years of sampling at most, but each provides a wealth of contemporaneous data at the regional scale.

This technical report provides summary data for nine distinct US JGOFS and international JGOFS sites. Four of these are point-locations where long-term time-series data have been collected: the Bermuda Atlantic Time-series Study (BATS); the Hawaiian Ocean Time-series (HOT); KERFIX, the French JGOFS site; and Station P, the Canadian JGOFS time-series station. The remaining sites are U.S. JGOFS Process Study sites: Arabian Sea; Equatorial Pacific (EqPac); North Atlantic Bloom Experiment (NABE); and the Antarctic Environment and Southern Ocean Survey (AESOPS) (which includes two distinct regions:

the Ross Sea and Antarctic Polar Front Zone). To provide an easy to use, practical presentation of data from each of these sites, both depth-profile data and the mixed layer average for each variable are determined on a cast-by-cast basis. Where possible, data are also standardized to common units. These data are available through the Data Support Section of the National Center for Atmospheric Research (<http://dss.ucar.edu/datasets/>) as dataset 259.0.

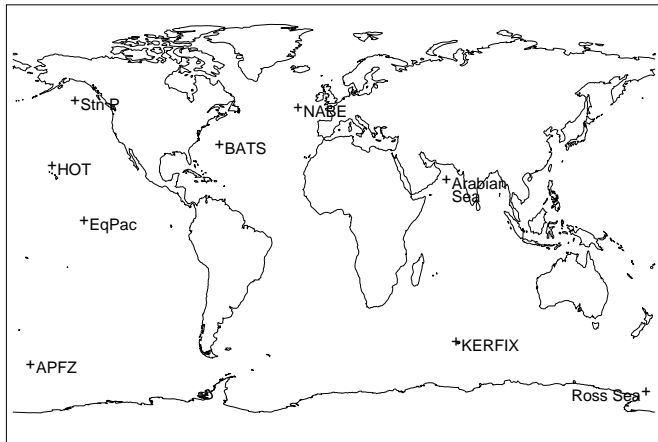


Fig. 1. Location of time-series stations and process study sites included in this report.

## 2. BASIC METHODOLOGY

### 2.1 DETERMINATION OF MIXED LAYER DEPTH

The mixed layer is that part of the surface ocean that is turbulently mixed by atmospheric processes (predominantly wind). A number of criteria are used as a measure of the mixed layer depth (MLD), based on temperature, density, or salinity. The reader is referred to You (1995, Appendix) for a concise summary of these criteria.

The criterion used in a particular oceanographic region depends on (1) which variables were measured (i.e., temperature, salinity, or both); (2) whether regional winds homogenize water properties to sufficient depths; and (3) the application (e.g. high frequency variability versus seasonal cycle). For example, the single most commonly used criterion is the depth where potential density ( $\sigma_\theta$ ) is  $0.125 \text{ kg m}^{-3}$  greater than the surface value. In most regions other than cold, polar environments, this corresponds closely with depth where temperature is  $0.05^\circ\text{C}$  cooler than at the surface (for example, Monterey and Levitus 1997, provide two values of MLDs for the world oceans using each of these criteria).



These two criteria work well in mid-latitudes, where strong winds result in deep, well-mixed surface layers, so that the MLD coincides with both the thermocline and pycnocline.

However, detecting MLD in both low latitude and high latitude oceans is more problematic. At high latitudes, the MLD is often difficult to detect, primarily because the vertical temperature gradient is extremely weak. At low latitudes, light winds result in insufficient mixing, so that the thermocline and pycnocline occur at significantly different depths. This is clearly the case in the EqPac study, where Wilf Gardner noted:

*“A year ago we submitted to the JGOFS EqPac data base the mixed-layer depths based on a density increase of 0.03 density units from surface values. This was the best fit for MLD compared to looking at the density profiles and noting the depth of the first minor break in slope of  $\sigma_\theta$ . This procedure was very sensitive to identifying the slight stratification that resulted from day-time solar heating, especially in regions of colder upwelled water. This also lead to maximizing the diel variation in MLD. We still believe this is the best criterion for investigating short time-scale processes in surface waters. For investigating longer-term processes, however, one might want to consider the MLD based on the commonly used Levitus standard of an increase of 0.125 density units from the surface value.”* (Note to EqPac Users: June 22, 1995, [http://usjgoofs.whoi.edu/PI-NOTES/Gardner\\_mixed.html](http://usjgoofs.whoi.edu/PI-NOTES/Gardner_mixed.html))

Based on the above observations, we have chosen one or several site-specific criteria to determine MLDs for the JGOFS sites:

Location	Latitude	Longitude	Reported Criteria	Used in Calculations
Station P	50°N	145°W	$\sigma_\theta=0.03$ & $0.125 \text{ kg m}^{-3}$	$\sigma_\theta=0.125 \text{ kg m}^{-3}$
BATS	31°40'N	64°10'W	$\sigma_\theta=0.03$ & $0.125 \text{ kg m}^{-3}$ pot. temp=0.1 & $0.5^\circ\text{C}$ † temp=0.1 & $0.5^\circ\text{C}$ †	$\sigma_\theta=0.125 \text{ kg m}^{-3}$ pot. temp= $0.5^\circ\text{C}$ temp= $0.5^\circ\text{C}$
HOT	22°45'N	138°W	$\sigma_\theta=0.03$ & $0.125 \text{ kg m}^{-3}$	$\sigma_\theta=0.125 \text{ kg m}^{-3}$
KERFIX	50°40'S	68°25'E	$\sigma_\theta=0.01$ & $0.02 \text{ kg m}^{-3}$	$\sigma_\theta=0.02 \text{ kg m}^{-3}$
Arabian Sea	10-23°N	57-69°E	$\sigma_\theta=0.03$ & $0.125 \text{ kg m}^{-3}$	$\sigma_\theta=0.125 \text{ kg m}^{-3}$
EqPac	12°N-12°S	140°W	$\sigma_\theta=0.03$ & $0.125 \text{ kg m}^{-3}$	$\sigma_\theta=0.125 \text{ kg m}^{-3}$
APFZ	55-65°S	170°E-170°W	$\sigma_\theta=0.01$ & $0.02 \text{ kg m}^{-3}$	$\sigma_\theta=0.02 \text{ kg m}^{-3}$
Ross Sea	70-73°S	170°E-170°W	$\sigma_\theta=0.01$ & $0.02 \text{ kg m}^{-3}$	$\sigma_\theta=0.02 \text{ kg m}^{-3}$
NABE	47°N	20°W	$\sigma_\theta=0.05$ & $0.125 \text{ kg m}^{-3}$	$\sigma_\theta=0.05 \text{ kg m}^{-3}$

†  $0.1^\circ\text{C}$  is usually equivalent to  $\sigma_\theta = 0.03 \text{ kg m}^{-3}$ ; and  $0.5^\circ\text{C}$  is usually equivalent to  $\sigma_\theta = 0.125 \text{ kg m}^{-3}$ .

The calculated MLDs for each site are illustrated in Figure 2. When possible, bottle data are used to calculate MLD. However, where bottle data are sparse, CTD data are used.

## Mixed Layer Depth

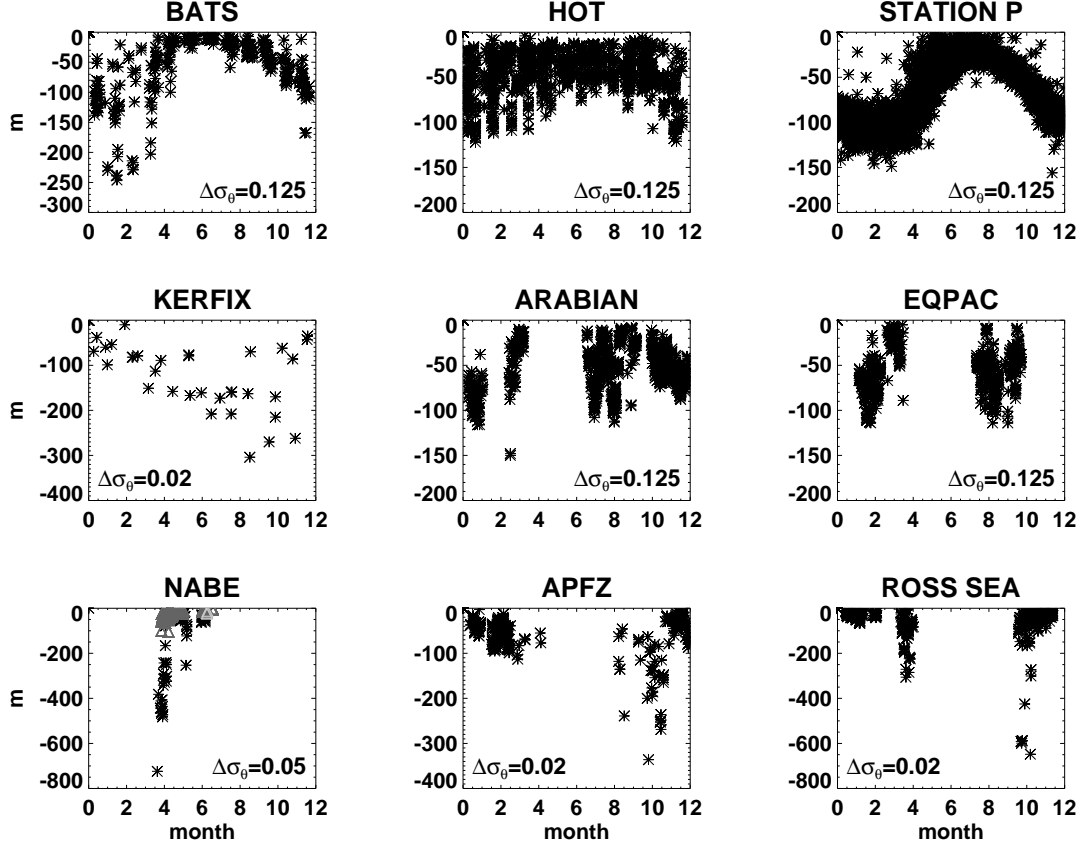


Fig. 2. Composite seasonal cycle of MLDs for each site. Criterion used for determining MLD is also shown. Data from BOFS (triangles) and PRIME (diamonds) taken within the NABE region are also shown. Note change in scale across sites.

MLDs for all stations are reported uniformly according to the following format:

Field	Description	Units
event	event number for cast (for ID purposes; often derived from date and cast no.)	number
date	date	YYMMDD
yday	year day	decimal days
lat	latitude	decimal degrees
lon	longitude	decimal degrees
mld#1	MLD according to criterion #1 (usually: 0.03 $\sigma_\theta$ for low-mid latitudes, 0.02 $\sigma_\theta$ for high latitudes)	m
mld#2	MLD according to criterion #2 (usually: 0.125 $\sigma_\theta$ for low-mid latitudes, 0.05 $\sigma_\theta$ for high latitudes)	m
st_depth	shallowest depth of profile	m

## 2.2 AVERAGING METHOD

Using the MLDs, we determined mixed layer averages of physical, chemical, and biological variables for each cast. Most physical water properties are fairly uniform or exhibit only weak gradients within the mixed layer (by definition). But this is not true for all biogeochemical properties, in particular primary production, which varies strongly with depth (light). The average mixed layer concentrations of variables are determined using the trapezoidal technique (Figure 3). Mixed layer averages for most of the measured bottle parameters are provided for each study site, along with mixed layer averages of primary production, and pigments. Also provided are the number of samples used to calculate the average.

## 2.3 PRIMARY PRODUCTION

Nearly every site reported primary production differently, though most sites used similar methodologies (note that no primary production data are available from Station P or KERFIX). We report primary production averaged over the mixed layer, in units standardized as  $\text{mmolC m}^{-3} \text{d}^{-1}$ . To obtain the integrated production for the mixed layer, one must multiply the average mixed layer production by the mixed layer depth, which yields production in  $\text{mmolC m}^{-2} \text{d}^{-1}$ .

All primary production measurements at the sites included in this report were performed using  $^{14}\text{C}$  incubations. However, primary production estimates are somewhat problematic.

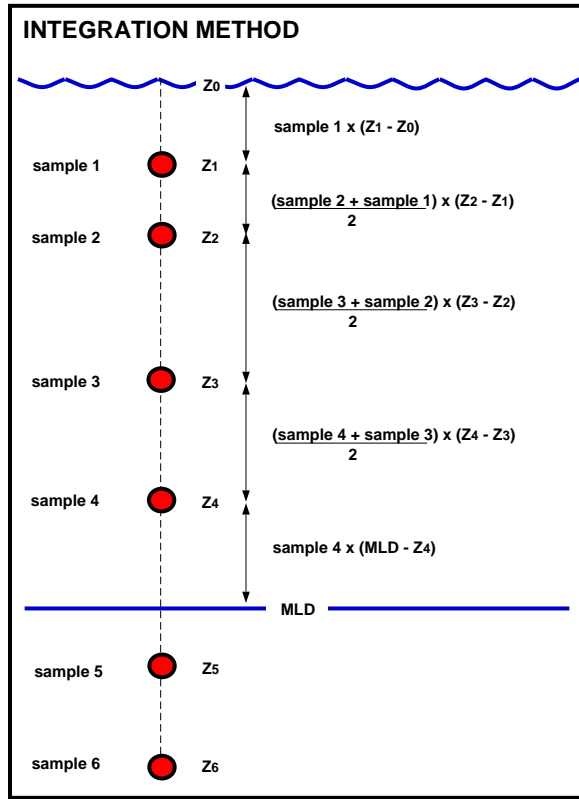


Fig. 3. Trapezoidal method for integrating variables within the mixed layer.

The conceptual and methodological problems of  $^{14}\text{C}$  incubations (e.g. assumption that  $^{14}\text{C}$ -DOC is negligible; problems of adsorption of  $^{14}\text{C}$ -DOC on filters) have been addressed by Karl et al. (1998), and the reader is referred to this paper for further information. Users of these data are also urged to consider the different field methodologies between sites, and how the incubations were originally reported.

Most study sites followed the primary production methodology outlined in the U.S. JGOFS Sampling and Analytical Protocols (see Barber 1993). This procedure uses in situ incubations of companion light and dark bottles at an array of depths within the euphotic zone, (surface to 0.1%  $I_0$  light depth); and primary production is usually determined as the light bottle less the dark bottle production (to account for non-photoautotrophic carbon fixation and adsorption). There are subtle differences between sites, however. Some studies conducted incubations from dawn to dusk, while others from dawn to dawn. Also, some experiments were conducted “on deck” in the ship laboratory, using water baths and arti-

ficial lighting to simulate those of the in situ array. And finally, some sites report primary production as daily rate while others report rate per incubation period. The following table outlines these differences between sites.

Site	Type	Duration	Original Units
BATS	in-situ	dawn-dusk	$\text{mgC m}^{-3} \text{ d}^{-1}$
HOT	in-situ and/or on-deck	dawn-dusk	$\text{mgC m}^{-3}$
NABE	in-situ	dawn-dusk	$\mu\text{mol C L}^{-1} \text{ time\_inc}^{-1}$
		dawn-dawn†	
EqPac	in-situ	dawn-dawn	$\text{mgC m}^{-3} \text{ d}^{-1}$
Arabian Sea	in-situ	dawn-dusk	$\text{mgC m}^{-3}$
		dawn-dawn†	
Southern Ocean	in-situ and/or on-deck	dawn-dawn	$\text{mmolC m}^{-3} \text{ d}^{-1}$

† Dusk-dawn portion of incubation was performed by placing bottles retrieved at dusk into an on-deck incubator for the dusk-dawn period.

The daily  $^{14}\text{C}$  primary production rates are usually based on 24 hour incubations, but at both HOT and BATS, incubations were conducted dawn-dusk. The BATS primary production data are reported as  $\text{mgC m}^{-3} \text{ d}^{-1}$ , but at HOT, it is reported as  $\text{mgC m}^{-3} \text{ incubation\_period}^{-1}$ . According to D. Karl (pers. comm.), “Since primary production does not occur at night (at least not the light dependent kind) ... per light day production estimates are effectively per day, once one deals with what to do about the dark bottles...” When using dawn-dusk incubations, a logical step might be to double the dark respiration values (to account for respiration both at day and night), but this is not always done. For this report HOT primary production rates are standardized to daily rates by merely subtracting the dark bottle value from the light bottle value.

Two sites, NABE and the Arabian Sea, provided concurrent data for both dawn-dusk and dawn-dawn incubations. At NABE, the difference in these estimates was reported as less than 15% (Knudson et al. 1998; Fig. 4). Similarly, a comparison of Arabian Sea 12 hour and 24 hour incubations shows that the 24 hour estimates are consistently lower by about 15–20%, which reflects the additional night-time respiratory loss.

Another difference between the primary production estimates is whether they were performed in situ or on-deck. Both in situ and on-deck incubations were performed in the Ross Sea. The on-deck incubations used the Morel optical model (Morel 1988; Barber et al. 1997) to estimate light intensity for the simulated incubation depths. A comparison of the two (Fig. 5) reveals that the methods produce results with considerable differences ( $r^2 = 0.73$ ), with on-deck incubations generally resulting in higher estimates.

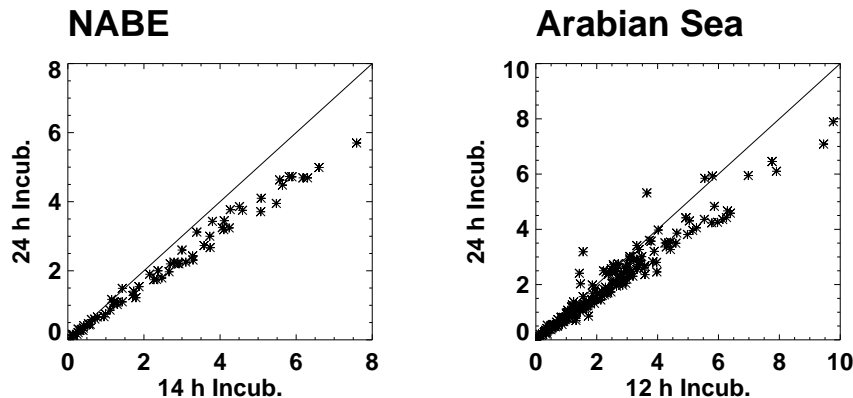


Fig. 4. Comparison of dawn-dawn (24 h) versus dawn-dusk (12 and 14 h) incubations at two sites, NABE and the Arabian Sea. All data are reported as  $\text{mmolC m}^{-3} \text{incubation\_period}^{-1}$ .

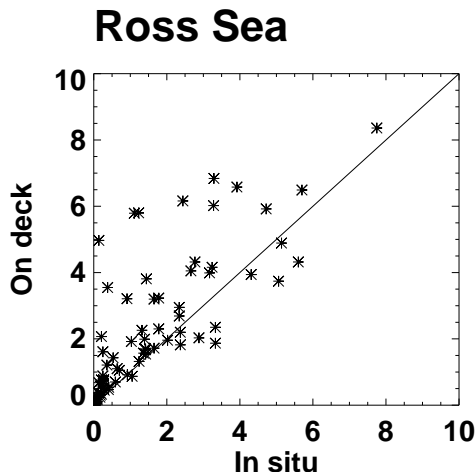


Fig. 5. Comparison of in situ versus on deck incubations at the Ross Sea site. All data are reported as  $\text{mmolC m}^{-3} \text{d}^{-1}$ .

## 2.4 UNIT CONVERSIONS

The tables in this report list for each variable both the original reported units, and the converted standardized units. For example, most of the time-series sites reported concentrations in  $\text{mg kg}^{-1}$  or  $\mu\text{mol kg}^{-1}$ , while the process study sites reported concentrations in  $\text{mg L}^{-1}$  or  $\mu\text{mol L}^{-1}$ . We adapted a standard of reporting concentrations in reference to water mass, and therefore report most parameters as concentration per kg (usually  $\mu\text{mol kg}^{-1}$ ). When converting from  $\text{L}^{-1}$  to  $\text{kg}^{-1}$ , the measured density of the same water sample is used; when not available, a representative density value for a particular site (e.g.  $1026 \text{ kg m}^{-3}$ ,  $1027 \text{ kg m}^{-3}$ ) is used. Pigment concentrations (e.g.,  $\mu\text{g L}^{-1}$  or  $\text{mg m}^{-3}$ ) are reported as weight per unit volume (usually  $\text{mg m}^{-3}$ ), primary production as moles per unit volume ( $\text{mmolC m}^{-3} \text{d}^{-1}$ ), and bacteria counts as  $\text{cells} \times 10^8 \text{ L}^{-1}$ , in agreement with the

most commonly used convention for these parameters.

### 3. RESULTS

The results of the data compilation for each site are reported using the following format:

A. Site description

B. Site-specific methodology

C. Description of reported variables

Each variable at a site is presented as two files: the mixed layer averages (\*.dat), and the discrete profile data (\*.pfl). The file naming convention generally follows the format of *site\_type\_VAR\_crit.dat* and *site\_type\_VAR.pfl* where:

<i>site</i>	study site (bats=BATS, arab=ARABIAN SEA, etc)
<i>type</i>	type of data (b=bottle, pp=primary production, etc)
<i>VAR</i>	variable analyzed (in caps unless derived from data)
<i>crit</i>	criterion used for mixed layer calculation (Pot_T=potential temperature, Sig_th= $\sigma_\theta$ , Temp=temperature) <i>Note that only the BATS and KERFIX sites report data using different MLD criteria, so this only exists in files for those sites</i>

The format for all mixed layer average files is as follows:

event	event number for the cast
date	date as YYMMDD
yday	year day as DDD.dd
lat	latitude
lon	longitude
mld	mixed layer depth
var_surf	surface value
var_avg	average mixed layer value as integrated total/mld
cnt	no. of samples in the mixed layer

### 3.1 TIME SERIES STATIONS

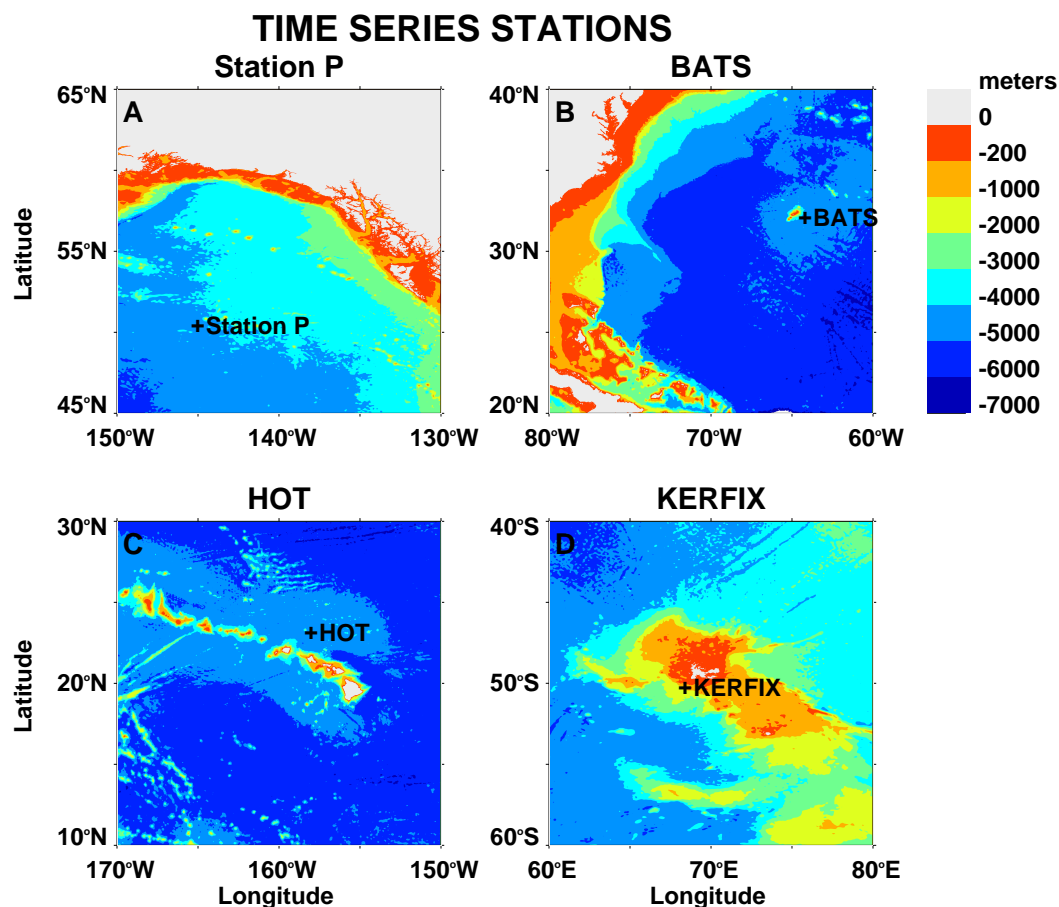


Fig. 6. Time-series stations shown in relation to the surrounding topography (topography extracted from 2-minute topography of Smith and Sandwell 1997).

#### 3.1.1 Station P (Station PAPA) (Fig. 6a)

Nominal Location:	50°N 145°W
Dates of Analysis:	1959 through 1995
Data Source:	Robin Brown Institute of Ocean Sciences P.O. Box 6000 Sidney, B.C. V8L 6B2 CANADA

Station P is essentially subarctic, and is characterized by high nitrate low chlorophyll (HNLC) waters and iron deficiency. The following excerpt, derived from the 1998 Coastal Salinity Workshop report (Woody et al. 1999) held in Hampton, Virginia, il-



illustrates the importance of this time-series station in detecting long-term trends in ocean physics/chemistry.

*“In the past decade many regions in the world’s oceans have experienced a decline in salinity and accompanying changes in the mixed layer depth. This is particularly true for the North Pacific where the mixed layer depth at Ocean Station P (50°N, 145°W) has decreased at a rate of about 63 m/century. Elsewhere in the coastal North Pacific, the mixed layer depth has been decreasing at a rate of about 32 m/century.”*

### **Site-specific Methodology**

None.

### **Description of reported variables**

#### *Station P bottle data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
stnp_b_TEMP	temperature	°C	°C
stnp_b_SAL	salinity	PSS	PSS
stnp_b_NIT	nitrate	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
stnp_b_PHO	phosphate	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
stnp_b_SIL	silicate	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
stnp_b_OXY	diss. oxygen	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$

#### *Station P chlorophyll and zooplankton biomass data*

Surface chlorophyll *a* data from Station P were originally compiled by Bruce Frost, who reported all chlorophyll measurements for a given year and day. We matched these data with average MLD for a given day.

Net zooplankton biomass data, also compiled by Bruce Frost, are mean monthly estimates of zooplankton biomass based on the data in Fulton (1980). Frost noted that prior to July 1966 samples were collected with a Norpac net; and after July 1966, with an S4 net. Based on comparisons of coincident data using both nets, Frost increased the original Norpac biomass values by a factor of 1.7705. This data set also included interpolated values for months where no data were available. We eliminated the interpolated values

from the data provided in this report.

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
stnp_chl_SURFCHLA	surf. chlorophyll <i>a</i>	mg m <sup>-3</sup>	mg m <sup>-3</sup>
stnp_zoo_ZOO	net zooplankton biomass	mg wet wt m <sup>-3</sup>	mg wet wt m <sup>-3</sup>

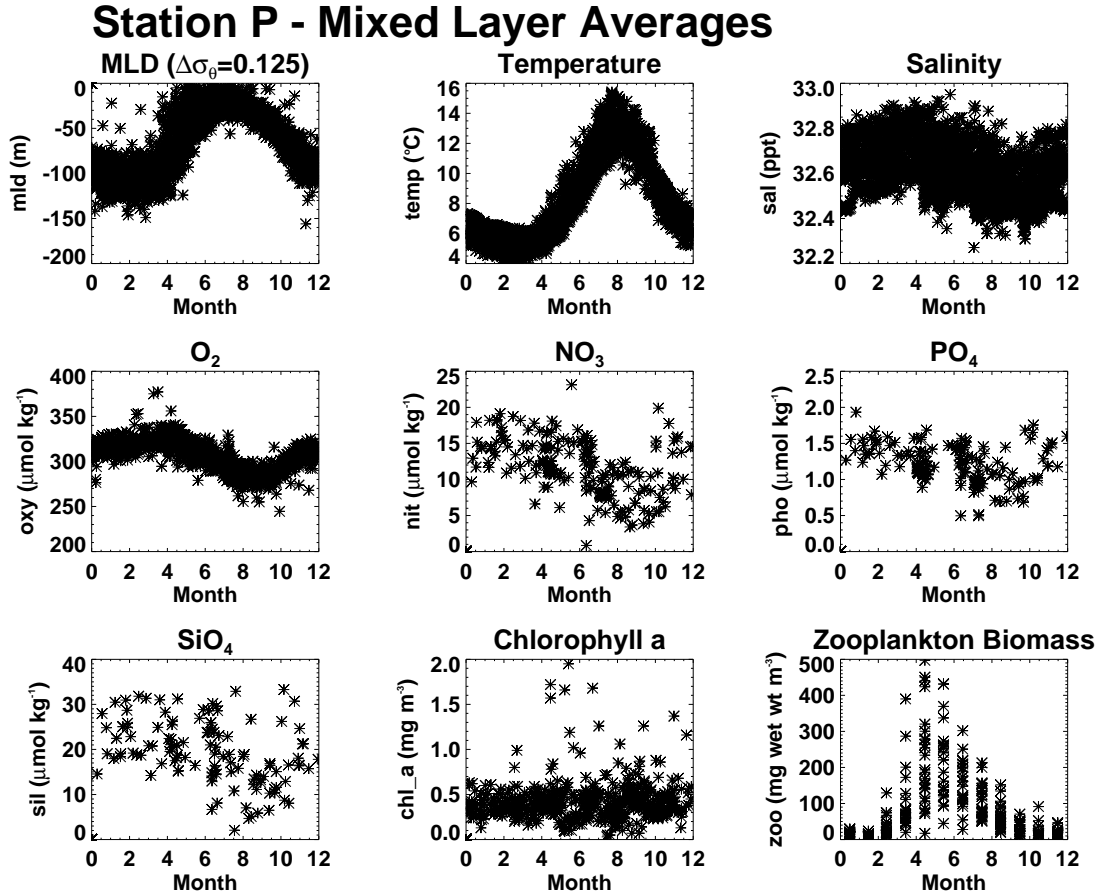


Fig. 7. Mixed layer averages for selected variables from the Station P site.

### 3.1.2 Bermuda Atlantic Time-series Study - BATS (Fig. 6b)

#### ***Site Description***

Nominal Location: 31°40'N, 64°10'W

Dates of Coverage: BATS CORE data are available beginning in Oct 1988

Dates of Analysis: Oct 1988 – Dec 31 1999

Data Source: <http://www.bbsr.edu/users/ctd/>

#### ***Site-specific Methodology***

For the BATS site, three different methods are used to calculate MLD:

- (1) depth where temperature was 0.5°C cooler than at the surface (*Temp*);
- (2) depth where potential temperature was 0.5°C cooler than at the surface (*Pot\_T*);
- (3) depth where  $\sigma_\theta$  was 0.125 greater than at the surface (*Sig\_th*).

*Method* in filenames reflect the method used. Each of these produced nearly identical results, but temperature measurements were available for all casts, and using this criterion produces more data points than using either potential temperature or  $\sigma_\theta$ .

## Description of reported variables

### BATS bottle data

FILENAME†	VARIABLE	ORIG UNITS	FINAL UNITS
bats_b_MLD_method	mixed layer depth	m	m
bats_b_T_method	temperature (bottle)	°C	°C
bats_b_SAL_method	salinity	PSS	PSS
bats_b_ANOM1_method	O <sub>2</sub> anomaly #1	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_ANOM2_method	O <sub>2</sub> anomaly #2	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_NO2_1_method	nitrite	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_NO3_1_method	nitrate	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_PO4_1_method	phosphate	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_O2_1_method	oxygen	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_O2_2_method	oxygen	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_SI_1_method	silicate	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_POC_method	particulate organic carbon	$\mu\text{g kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_TOC_method	total organic carbon	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_doc_method‡	dissolved organic carbon	n/a	$\mu\text{mol kg}^{-1}$
bats_b_PON_method	particulate organic nitrogen	$\mu\text{g kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_TON_method	total organic nitrogen	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_don_method§	dissolved organic nitrogen	n/a	$\mu\text{mol kg}^{-1}$
bats_b_cton_method¶	C to N ratio of particulate organic matter	n/a	ratio
bats_b_TCO2_method	total carbon dioxide	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
bats_b_CHL_method	Turner chlorophyll <i>a</i>	$\mu\text{g kg}^{-1}$	$\text{mg m}^{-3}$
bats_b_PHAEO_method	Turner phaeopigments	$\mu\text{g kg}^{-1}$	$\text{mg m}^{-3}$
bats_b_BACT_method	bacteria	$\text{cells} \times 10^8 \text{ kg}^{-1}$	$\text{cells} \times 10^8 \text{ L}^{-1}$

† *method* in filenames reflect which mixed layer criterion is used: *Temp* = temperature difference of 0.5°C ; *Pot\_T* = potential temperature difference of 0.5°C ; *Sig\_th* = density difference of 0.125 $\sigma_\theta$

‡ Calculated as TOC–POC (sample TOC – sample POC; then those values averaged).

§ Calculated as TON–PON (sample TON – sample PON; then those values averaged).

¶ Calculated as ratio of molar concentration of POC to molar concentration of PON.

*BATS HPLC pigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
bats_hplc_P1_method	chlorophyll $c_3$	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P2_method	chlorophyllide $a$	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P3_method	chlorophyll $c_1 + c_2$ & chlorophyll Mg 3,8DVP $a_5$	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P4_method	peridinin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P5_method	19'-butanoyloxyfucoxanthin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P6_method	fucoxanthin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P7_method	19'-hexanoyloxyfucoxanthin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P8_method	prasincoxanthin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P9_method	diadinoxanthin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P10_method	alloxanthin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P11_method	diatoxanthin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P12_method	zeaxanthin+lutein	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P13_method	chlorophyll $b$	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P14_method	chlorophyll $a$	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P15_method	$\alpha + \beta$ -carotene	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P18_method	lutein	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P19_method	zeaxanthin	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P20_method	$\alpha$ -carotene	ng kg <sup>-1</sup>	mg m <sup>-3</sup>
bats_hplc_P21_method	$\beta$ -carotene	ng kg <sup>-1</sup>	mg m <sup>-3</sup>

*BATS primary production data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
hplc			
bats_pp_LT1_method	<sup>14</sup> C prim prod light btl #1	mgC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
bats_pp_LT2_method	<sup>14</sup> C prim prod light btl #2	mgC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
bats_pp_LT3_method	<sup>14</sup> C prim prod light btl #3	mgC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
bats_pp_DARK_method	<sup>14</sup> C prim prod dark btl	mgC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
bats_pp_T0_method	<sup>14</sup> C prim prod time 0	mgC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
bats_pp_PP_method	<sup>14</sup> C prim prod light - dark	mgC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
bats_pp_THY1_method	Bact gr rate, <sup>3</sup> H-thym. #1	pmol L <sup>-1</sup> h <sup>-1</sup>	pmol L <sup>-1</sup> h <sup>-1</sup>
bats_pp_THY2_method	Bact gr rate, <sup>3</sup> H-thym. #2	pmol L <sup>-1</sup> h <sup>-1</sup>	pmol L <sup>-1</sup> h <sup>-1</sup>
bats_pp_THY3_method	Bact gr rate, <sup>3</sup> H-thym. #3	pmol L <sup>-1</sup> h <sup>-1</sup>	pmol L <sup>-1</sup> h <sup>-1</sup>
bats_pp_THY_method	Mean bacterial growth rate	pmol L <sup>-1</sup> h <sup>-1</sup>	pmol L <sup>-1</sup> h <sup>-1</sup>

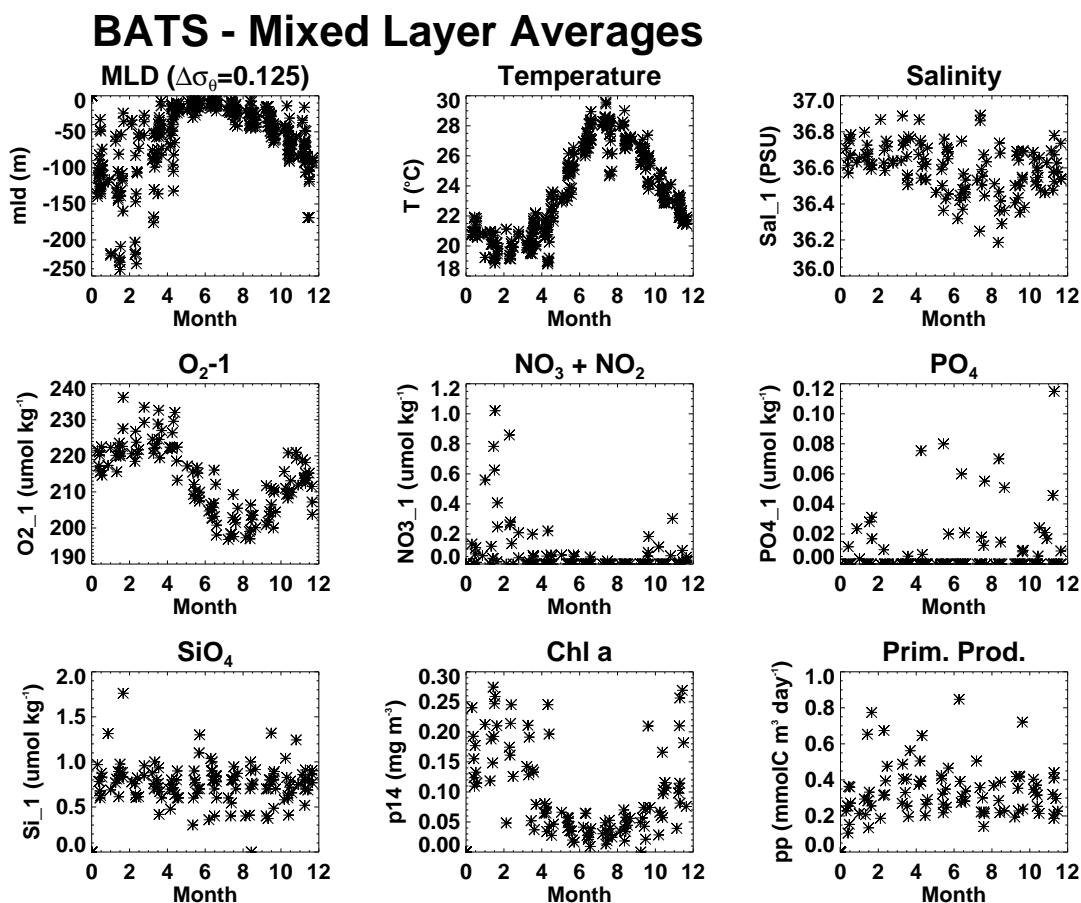


Fig. 8. Mixed layer averages for selected variables from the BATS site.

### 3.1.3 Hawaiian Ocean Times-series - HOT (Fig. 6c)

Nominal Location: 22°45'N, 158°W (Station ALOHA)  
 Dates of Coverage: HOT ALOHA data are available beginning in Oct 1988  
 Dates of Analysis: Oct 1988 – Dec 1998  
 Data Source: <http://hahana.soest.hawaii.edu/hot/hot-dogs/interface.html>

### Site-specific Methodology

None.

## Description of reported variables

### HOT bottle data

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
aloha_b_TEMP	temperature	°C	°C
aloha_b_SAL_CTD	CTD salinity	PSS-78	PSS-78
aloha_b_SAL_BOT	bottle salinity (Autosal)	PSS-78	PSS-78
aloha_b_OXY_BOT	bottle oxygen	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_DIC	dissolved inorganic carbon	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_PH	pH	pH units	pH units
aloha_b_ALK	alkalinity	$\mu\text{eq kg}^{-1}$	$\mu\text{eq kg}^{-1}$
aloha_b_PHOS	phosphate, $\text{PO}_4$	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_NIT	$\text{NO}_3 + \text{NO}_2$	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_SIL	silicate, $\text{SiO}_4$	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_DOP	dissolved organic phosphorus	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_DON	dissolved organic nitrogen	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_DOC	dissolved organic carbon	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_TDP	total dissolved phosphorus	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_TDN	total dissolved nitrogen	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_PC†	particulate carbon	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_PN	particulate nitrogen	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
aloha_b_cton‡	C to N ratio of particulate matter	n/a	ratio
aloha_b_PPH	particulate phosphorus	$\text{nmol kg}^{-1}$	$\text{nmol kg}^{-1}$
aloha_b_CHL	fluorometric chlorophyll	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$

† Note that particulate carbon includes both particulate organic carbon and particulate inorganic carbon.

‡ Calculated as ratio of molar concentration of PC to molar concentration of PN. Note that this differs from the normal calculation of C:N, which is determined as the molar ratio of POC to PON.

### *HOT HPLC pigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
aloha_hplc_CHL3	chlorophyll $c_3$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_CHL12	chlorophyll [ $c_1 + c_2$ ] & Mg 3,8 DVP4 $a_5$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_CHLPLUS	chlorophyll $c_1, c_2, c_3$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_PERID	peridinin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_BUT19	19'-butanoyloxyfucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_FUCO	fucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_HEX19	19'-hexanoyloxyfucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_PRASINO	prasincoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_DIADINO	diadinoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_ZEAXAN	zeaxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_CHLB	chlorophyll $b$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_CHLA	chlorophyll $a$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_CHLC4	chlorophyll $c_4$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_ACAR	$\alpha$ -carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_BCAR	$\beta$ -carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_CAROTEN	carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_CHLDA	chlorophyllide $a$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_VIOL	violaxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_LUTEIN	lutein	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_MVCHLA	monovinyl chlorophyll $a$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
aloha_hplc_DVCHLA	divinyl chlorophyll $a$	ng L <sup>-1</sup>	mg m <sup>-3</sup>

### *HOT primary production data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
aloha_pp_CHLA	chlorophyll $a$	mg m <sup>-3</sup>	mg m <sup>-3</sup>
aloha_pp_LT12	light 12	mg C m <sup>-3</sup> incub-per <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
aloha_pp_DK12	dark 12	mg C m <sup>-3</sup> incub-per <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
aloha_pp-pp†	prim. prod, (lt12-dk12)	mg C m <sup>-3</sup> incub-per <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>

Calculated as the LT12 value minus the DK12 value.

### *HOT bacteria data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
aloha_bact_HBACT	heterotrophic bacteria	cells×10 <sup>5</sup> ml <sup>-1</sup>	cells×10 <sup>8</sup> L <sup>-1</sup>
aloha_bact_PBACT	Prochlorococcus	cells×10 <sup>5</sup> ml <sup>-1</sup>	cells×10 <sup>8</sup> L <sup>-1</sup>
aloha_bact_SBACT	Synechococcus	cells×10 <sup>5</sup> ml <sup>-1</sup>	cells×10 <sup>8</sup> L <sup>-1</sup>
aloha_bact_EBACT	eukaryotes	cells×10 <sup>5</sup> ml <sup>-1</sup>	cells×10 <sup>8</sup> L <sup>-1</sup>



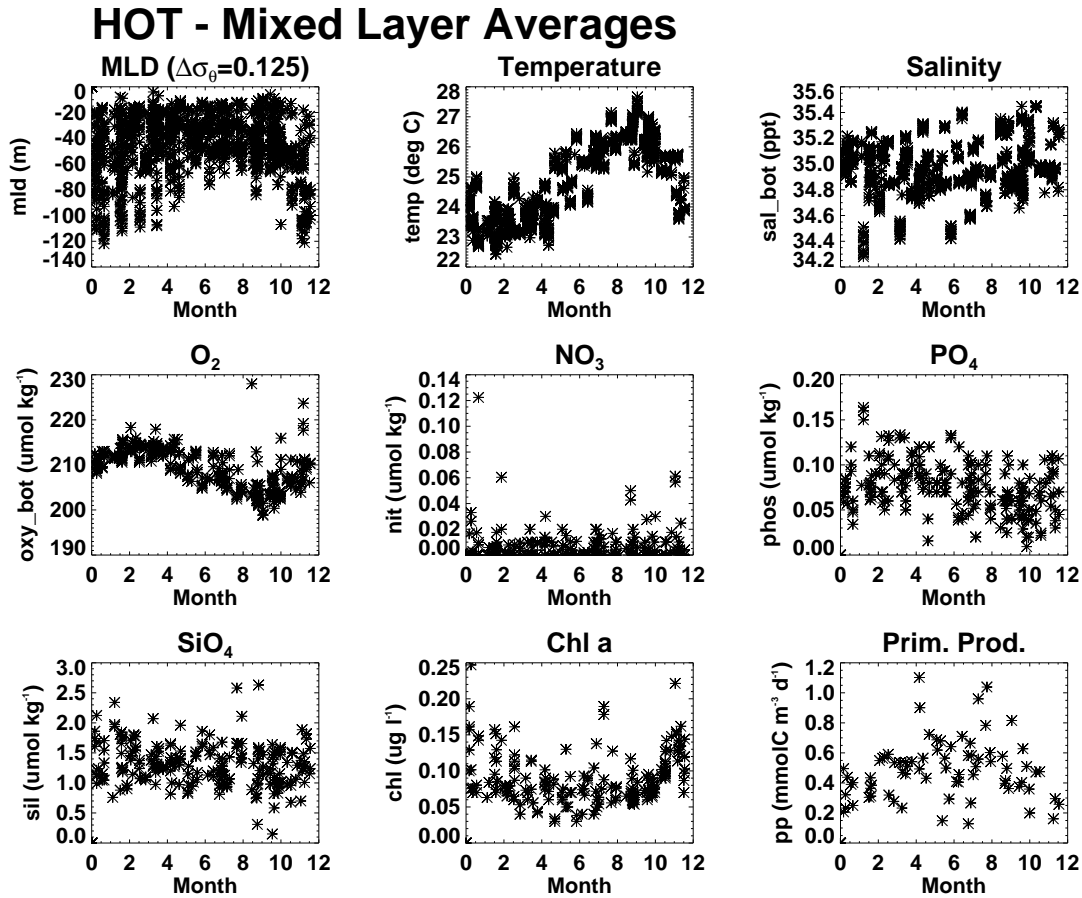


Fig. 9. Mixed layer averages for selected variables from the HOT ALOHA site.

#### 3.1.4 KERFIX (Fig. 6d)

Nominal Location:	50°40'S, 68°25'E, 100km SW of Kerguelen Island
Dates of Coverage:	near-monthly observations between 1990 and 1995.
Dates of Analysis:	Apr 1990 – Mar 1995
Data Source:	<a href="http://www.obs-vlfr.fr/jgofs/html/html/acces_base.html">http://www.obs-vlfr.fr/jgofs/html/html/acces_base.html</a>

## Site-specific Methodology

Two sets of data are provided for the KERFIX site. The first is based on MLDs calculated for each cast. These data are consistent with other sites in this report; however, there are numerous casts for which the MLD could not be calculated.

The second dataset is based on MLDs determined by Park et al. (1998). Park et al. provided MLD determinations for most months between Dec 1990 – Dec 1994, in three separate files: for year/month (*kerfixmld\_ymavg*), averaged monthly (*kerfixmld\_mavg*), and the root-mean-squared variability of monthly mld (*kerfixmld\_rms*). His year/month MLDs are used to calculate mixed layer averages of this alternative KERFIX data set. Park’s MLDs were determined using a sequence of interpolations: (1) linear interpolation of missing pressure values; (2) linear interpolation to regular depth intervals of 10 m; and finally, (3) linear interpolation to 15th of each month.

## Description of reported variables

### KERFIX bottle data

FILENAME†	VARIABLE	ORIG UNITS	FINAL UNITS
<i>kerfix_b_TEMP_method</i>	temperature	°C	°C
<i>kerfix_b_SAL_method</i>	salinity	PSS	PSS
<i>kerfix_b_OXY_method</i>	oxygen	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
<i>kerfix_b_NO3_method</i>	nitrate, $\text{NO}_3$	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
<i>kerfix_b_PO4_method</i>	phosphate, $\text{PO}_4$	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
<i>kerfix_b_SIO2_method</i>	silicate, $\text{SiO}_2$	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
<i>kerfix_b_NH4_method</i>	ammonium, $\text{NH}_4$	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
<i>kerfix_b_CHLA_method</i>	chlorophyll <i>a</i>	$\text{mg m}^{-3}$	$\text{mg m}^{-3}$
<i>kerfix_b_TCO2_method</i>	total $\text{CO}_2$	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
<i>kerfix_b_PCO2_method</i>	partial pressure of $\text{CO}_2$	$\mu\text{atm}$	$\mu\text{atm}$
<i>kerfix_b_ALK_method</i>	alkalinity	$\mu\text{eq kg}^{-1}$	$\mu\text{eq kg}^{-1}$

† *method* in filenames reflect which MLD determination is used: *botmld* = MLD determined from bottle data; *park* = MLD interpolated from Park et al. (1998) monthly values.

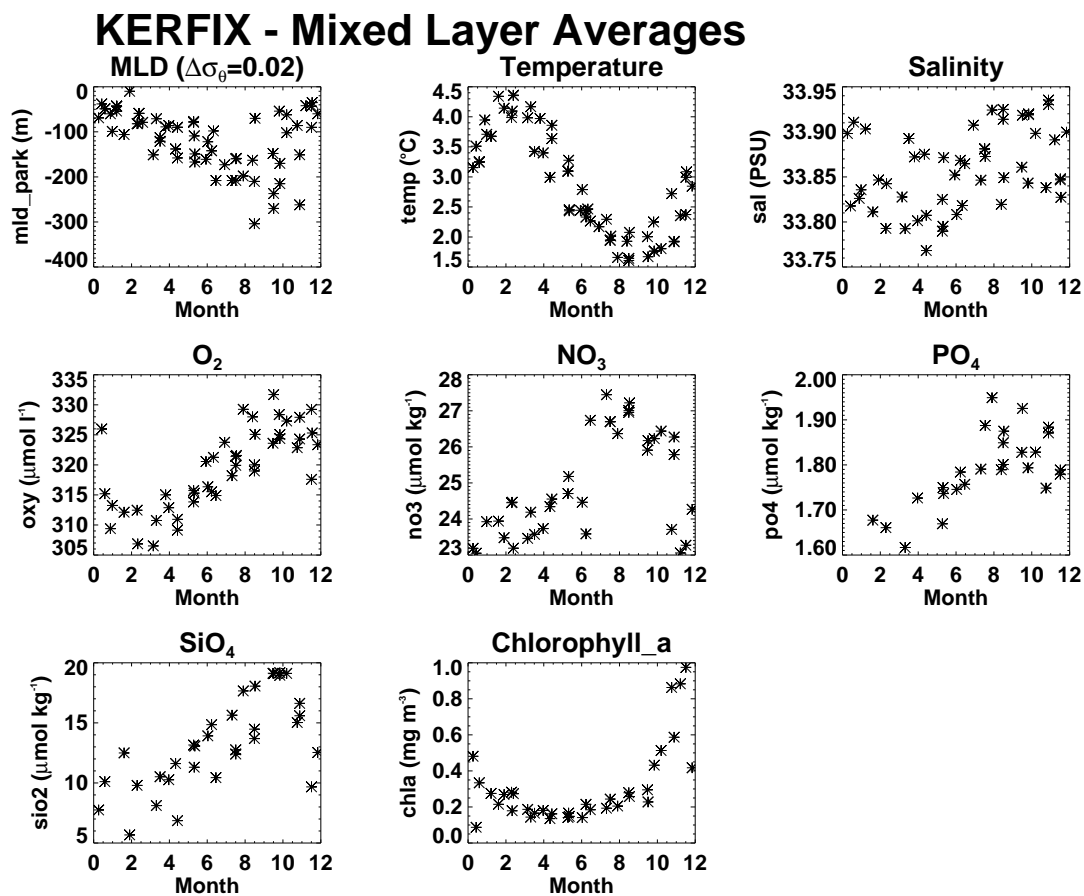


Fig. 10. Mixed layer averages for selected variables from the KERFIX site.

## 3.2 PROCESS STUDY SITES

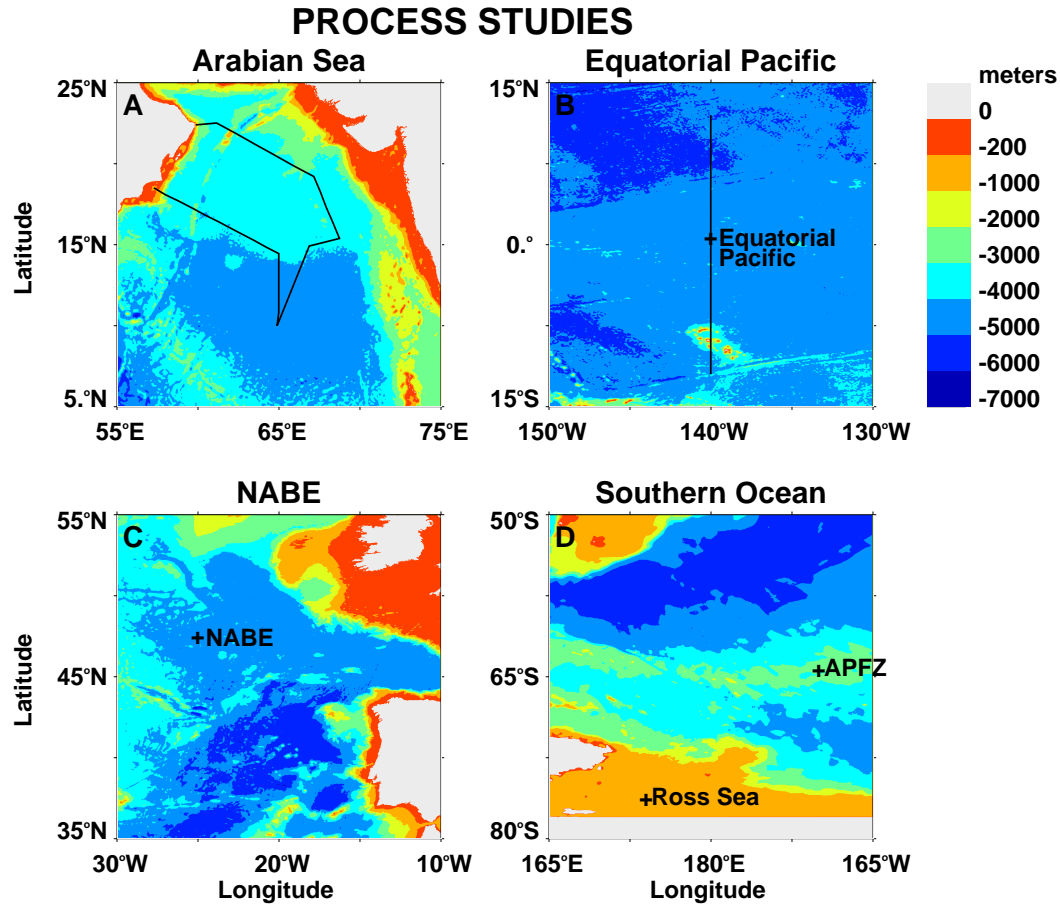


Fig. 11. Process study sites shown in relation to the surrounding topography (topography extracted from 2-minute topography of Smith and Sandwell 1997; except for Southern Ocean (11D) which was extracted from TerrainBase (Row et al. 1994, Row et al. 1995).

### 3.2.1 Arabian Sea (Fig. 11a)

Regional Coverage:	Arabian Sea from 10–23°N and 57–69°E
Dates of Coverage:	Sep 1994 – Dec 1995
Data Source:	<a href="http://usjgofs.whoi.edu/jg/dir/jgofs/arabian/">http://usjgofs.whoi.edu/jg/dir/jgofs/arabian/</a>

### ***Site-specific Methodology***

MLDs, determined for each CTD cast, were already available for most cruises, as provided by Wilford Gardner via the US JGOFS data system. Gardner provided four separate values based on: (a) 0.03 density increase relative to surface; (b) 0.1°C decrease relative to the surface, a nearly equivalent measure to (a); (c) 0.125 density increase relative to surface; and (d) 0.1°C decrease relative to the surface, a nearly equivalent measure to (c).

However, since MLDs were not available for all cruises, these are recalculated according to Gardner’s criteria (a) and (c) and using the CTD data from the US JGOFS data server. These “new” MLD files are provided as: *ttn-crno\_mld.dat* where *crno* is the Arabian Sea cruise number. The 0.125 density criterion is used to calculate mixed layer averages for Arabian Sea data.

### ***Description of reported variables***

*Arabian Sea bottle data:*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
arab_b_TEMP	from CTD	°C	°C
arab_b_SAL_CTD	CTD salinity	PSU	PSU
arab_b_SAL_BOT	bottle salinity (Autosal)	PSU	PSU
arab_b_O2_1	oxygen (Winkler)	ml L <sup>-1</sup>	ml L <sup>-1</sup>
arab_b_O2_2	oxygen (Winkler)	μmol kg <sup>-1</sup>	μmol kg <sup>-1</sup>
arab_b_O2_3	oxygen (Winkler)	μmol L <sup>-1</sup>	μmol L <sup>-1</sup>
arab_b_NO3	nitrate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
arab_b_PO4	reactive phosphorus	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
arab_b_SiO4	silicic acid/reactive silica	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
arab_b_NO2	nitrite	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
arab_b_NH4	ammonium	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>

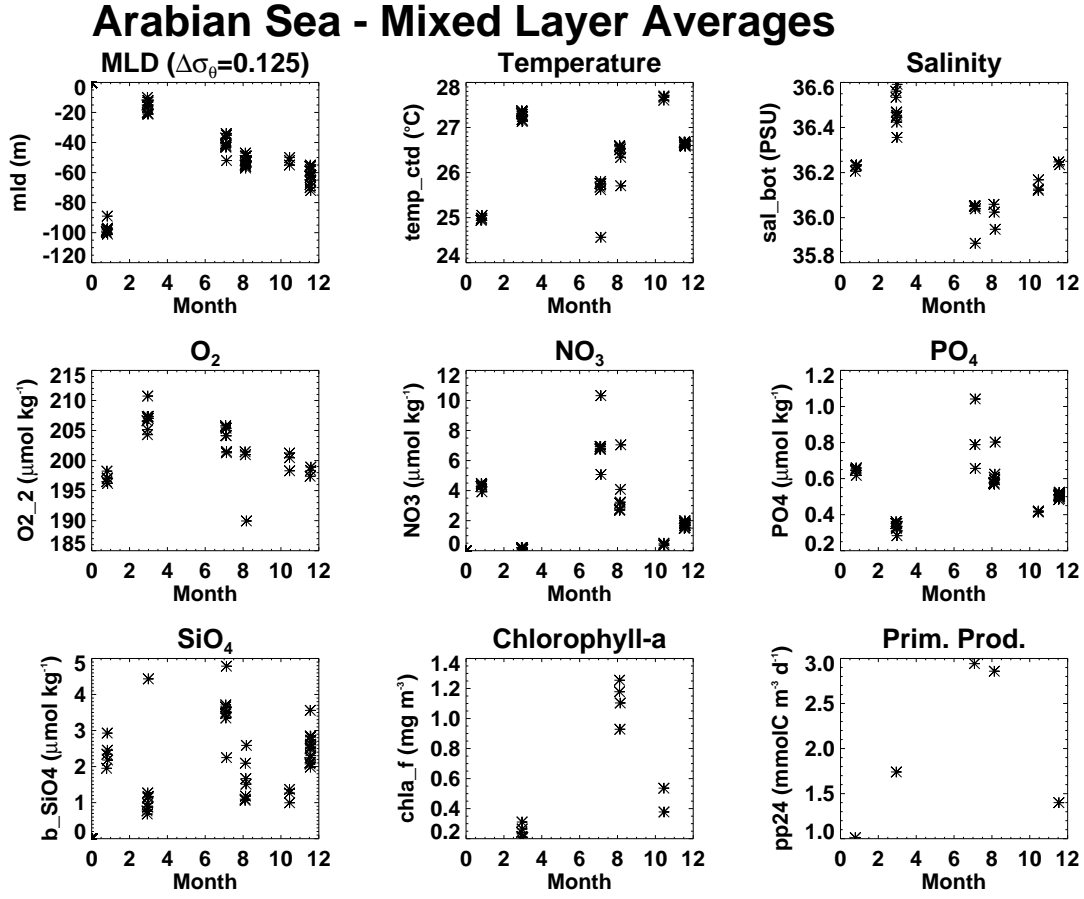


Fig. 12. Mixed layer averages for selected variables from the Arabian Sea Process Study. Because data cover a wide spatial area, shown here is a subset of data from the main data set, extracted for a  $1^{\circ}$  grid cell centered at  $16^{\circ}\text{N}$  and  $62^{\circ}\text{E}$ .

#### Arabian Sea primary production

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
arab_pp_CHL_A	chl <i>a</i> fluoro. method	$\text{mg m}^{-3}$	$\text{mg m}^{-3}$
arab_pp_PP12	prim. prod., C assim dawn-dusk (12hr)	$\text{mgC m}^{-3} \dagger$	$\text{mmolC m}^{-3} 12\text{h}^{-1}$
arab_pp_PB12	C assim. per unit chl <i>a</i> dawn-dusk (12hr)	$\text{mgC m}^{-3} \dagger$	$\text{mmolC mgChl}^{-1} 12\text{h}^{-1}$
arab_pp_PP24	prim. prod., C assim dawn-dawn (24hr)	$\text{mgC m}^{-3} \dagger$	$\text{mmolC m}^{-3} \text{ d}^{-1}$
arab_pp_PB24	C assim. per unit chl <i>a</i> dawn-dawn (24hr)	$\text{mgC m}^{-3} \dagger$	$\text{mmolC mgChl}^{-1} \text{ d}^{-1}$

$\dagger$  Stated units are in  $\text{mgC m}^{-3}$ ; the units for both of these values are actually  $\text{mgC mgChl}^{-1} \text{ d}^{-1}$ , with one day equivalent to either the 12 h or 24 h incubation.

*Arabian Sea chlorophyll-fluorometric data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
arab_df_CHLA_F	chlorophyll <i>a</i> fluorometric	mg m <sup>-3</sup>	mg m <sup>-3</sup>
arab_cf_PHAEO	phaeopigments	mg m <sup>-3</sup>	mg m <sup>-3</sup>

*Arabian Sea HPLC pigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
arab_hplc_CHL_A1	chlorophyll <i>a</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_A1p	chlorophyll <i>a'</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_A2	divinyl-chl <i>a</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_A_TOT	chl <i>a</i> <sub>1</sub> , chl <i>a</i> <sub>2</sub> , chl dea, chl <i>a'</i> <sub>1</sub> , & allomerized chl <i>a</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_B1	chlorophyll <i>b</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_B2	divinyl-chl <i>b</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_B_TOT	sum of chl <i>b</i> <sub>1</sub> and chl <i>b</i> <sub>2</sub>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_C	sum of chl <i>c</i> <sub>1</sub> , chl <i>c</i> <sub>2</sub> , & Mg 3,8 divinyl-pheoporphyrin <i>a</i> <sub>5</sub>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_C3	chlorophyll <i>c</i> <sub>3</sub>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_C4_1	phytolated chl <i>c</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHL_C4_2	phytolated chl <i>c</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHLIDE_A	chlorophyllide <i>a</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CHLIDE_B	chlorophyllide <i>b</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_PERIDININ	peridinin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_FUCOX	fucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_FUCOX_BUT	19'-butanoyloxyfucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_FUCOX_HEX	19'-hexanoyloxyfucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CIS_FUCOX	cis-fucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CIS_HEX	cis-19'-hexanoyloxy-fucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CAROTENE_A	$\alpha$ -carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CAROTENE_B	$\beta$ -carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_CAROTENE	$\alpha$ & $\beta$ -carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_ALLOX	alloxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_DIADINOX	diadinoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_DIATOX	diatoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_LUTEIN	lutein	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_NEOX	neoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_PRASINOX	prasincoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_VIOLAX	violaxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
arab_hplc_ZEAX	zeaxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>

### *Arabian Sea bacteria data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
arab_bact_BACT	count of bacteria cells	cells $\times 10^9$ L $^{-1}$	cells $\times 10^8$ L $^{-1}$

### *Arabian Sea particulate organic matter and dissolved carbon data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
arab_poc_POC	particulate organic carbon	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
arab_poc_PON	particulate organic nitrogen	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
arab_poc_CTON	C to N ratio of particulate organic matter	ratio	ratio
arab_toc_TOC	total organic carbon	$\mu\text{mol L}^{-1}$ , and $\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
arab_doc †	dissolved organic carbon	n/a	$\mu\text{mol kg}^{-1}$

† Calculated as difference between particulate organic carbon and total organic carbon

### **3.2.2 Equatorial Pacific - EqPac (Fig. 11b)**

Nominal Track:	N-S line, 12°S – 12°N, along 140°W
Dates of Coverage:	All cruises were conducted within the calendar year 1992.
Data Source:	<a href="http://usjgofs.whoi.edu/jg/dir/jgofs/eqpac/">http://usjgofs.whoi.edu/jg/dir/jgofs/eqpac/</a>

### ***Site-specific Methodology***

MLDs were provided by Wilf Gardner as two depths: where potential density was 0.03 kg m $^{-3}$  and 0.125 kg m $^{-3}$  greater than the surface. Dr. Gardner recommended using 0.03 kg m $^{-3}$  for short term determinations of MLD (e.g. diel variation) but that for longer-term processes, MLD based on the Levitus standard (0.125 kg m $^{-3}$  difference) may be more appropriate. For our calculations, we used the 0.125 kg m $^{-3}$  criterion.



## *Description of reported variables*

### *Equatorial Pacific bottle data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
eqpac_b_TEMP_CTD	CTD temperature	°C	°C
eqpac_b_SAL_CTD	CTD salinity	PSS	PSS
eqpac_b_SIGMA_T_CTD	CTD sigma-theta	kg m <sup>-3</sup>	kg m <sup>-3</sup>
eqpac_b_POTEMP_CTD	CTD potential temperature	°C	°C
eqpac_b_SIGMA_0_CTD	CTD potential density	kg m <sup>-3</sup>	kg m <sup>-3</sup>
eqpac_b_O2_BOT	dissolved oxygen (Winkler)	ml L <sup>-1</sup>	ml L <sup>-1</sup>
eqpac_b_O2	dissolved O <sub>2</sub> conc	μmol O <sub>2</sub> kg <sup>-1</sup>	μmol O <sub>2</sub> kg <sup>-1</sup>
eqpac_b_SAL_BOT	bottle salinity	PSS	PSS
eqpac_b_NH4	ammonium	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
eqpac_b_NO3	nitrate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
eqpac_b_NO2	nitrite	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
eqpac_b_PO4	phosphate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
eqpac_b_SIO4	silicate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
eqpac_b_UREA	urea	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
eqpac_b_CHL_A	chlorophyll <i>a</i>	μg L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_b_PHAEO	total phaeopigments	μg L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_b_FO_TO_FA	ratio of fluorometric reading before (Fo) and after (Fa) acidification	ratio	ratio

### *Equatorial Pacific primary production data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
eqpac_pp_CHL_A	chlorophyll <i>a</i>	mg m <sup>-3</sup>	mg m <sup>-3</sup>
eqpac_pp_PP24	primary production, C assimilation	mgC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
eqpac_pp_PB24	chl <i>a</i> specific production	mgC mgChl <sup>-1</sup> d <sup>-1</sup>	mmolC mgChl <sup>-1</sup> d <sup>-1</sup>

*Equatorial Pacific HPLC pigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
eqpac_hplc_CHL_C3	chlorophyll $c_3$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_CLID	chlorophyllide $a$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_CHL_C1_C2	chlorophylls $c_1$ and $c_2$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_PER	peridinin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_FUCOX_BUT	19'-butanoyloxyfucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_FUCOX	fucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_FUCOX_HEX	19'-hexanoyloxyfucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_PRAS	prasincoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_VIOL	violaxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_DIAD	diadinoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_ALLOX	alloxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_DIATOX	diatoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_ZEA	zeaxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_CHL_B	chlorophyll $b$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_ALLOA	allomerized chlorophyll $a$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_CHL_A	chlorophyll $a$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_C4	phytolated chl $c$ -like pigment	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_CHL_Ap	chlorophyll $a'$	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_CAROTENE_A	$\alpha$ -carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>
eqpac_hplc_CAROTENE_B	$\beta$ -carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>

*Equatorial Pacific bacteria data†*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
eqpac_bact_THY_INCORP	thymidine incorporation	pmol L <sup>-1</sup> hr <sup>-1</sup>	pmol L <sup>-1</sup> hr <sup>-1</sup>
eqpac_bact_LEU_INCORP	leucine incorporation	pmol L <sup>-1</sup> hr <sup>-1</sup>	pmol L <sup>-1</sup> hr <sup>-1</sup>
eqpac_bact_BACT	bacteria abundance	cells×10 <sup>6</sup> ml <sup>-1</sup> , cells×10 <sup>8</sup> L <sup>-1</sup>	cells×10 <sup>8</sup> L <sup>-1</sup>

† For samples from EqPac cruise tt007, Dave Kirchman noted: "... some samples that were clearly anomalous, but again we did not delete or change these. Some of these anomalous values may be due to problems with the CTD bottles, but others are clearly not. (see [http://usjgofs.whoi.edu/PI-NOTES/Kirchman\\_bact.html](http://usjgofs.whoi.edu/PI-NOTES/Kirchman_bact.html) for details).

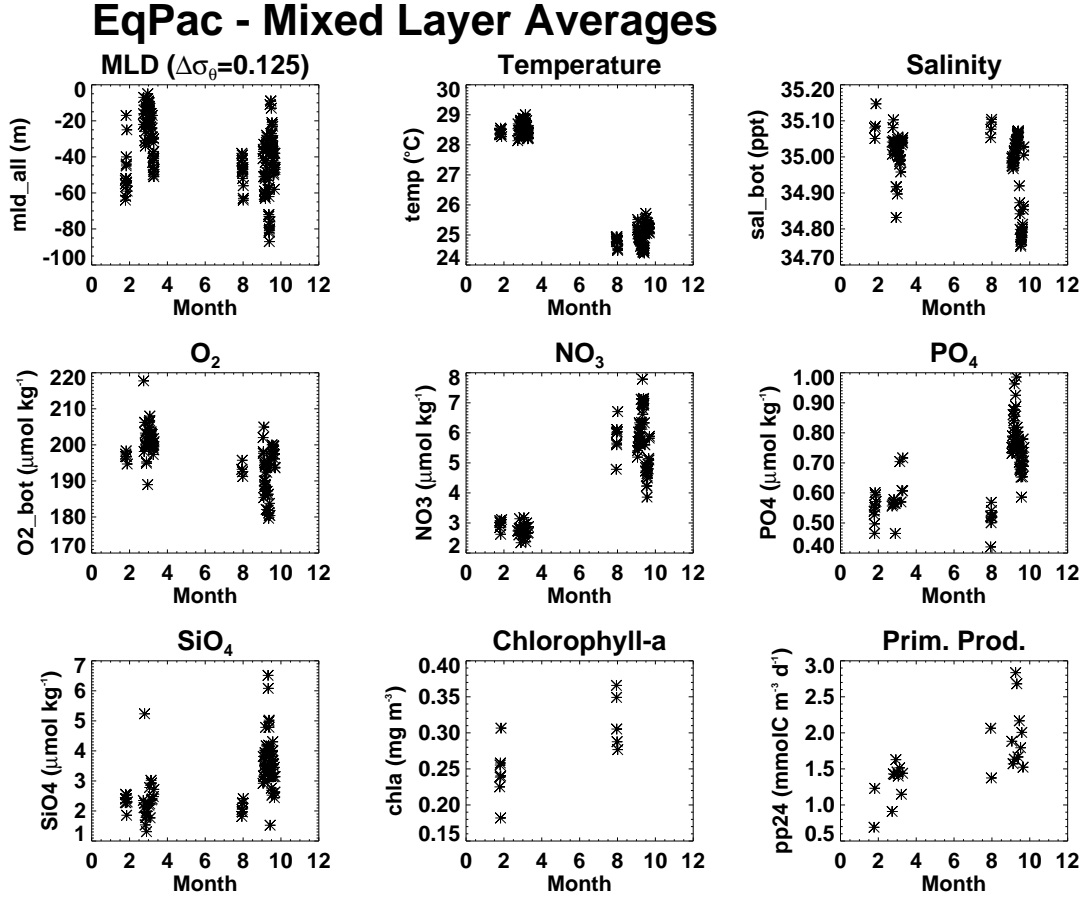


Fig. 13. Mixed layer averages for selected variables from Equatorial Pacific Process Study. Because data cover a wide spatial area, shown here is a subset of data from the main data set, extracted for a  $1^{\circ}$  grid cell centered at  $0^{\circ}$  and  $140^{\circ}\text{W}$ .

#### *Equatorial Pacific particulate and dissolved organic matter†*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
eqpac_poc_TPC	total particulate carbon	$\mu\text{g L}^{-1}, \mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
eqpac_poc_POC	particulate organic carbon	$\mu\text{g L}^{-1}, \mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
eqpac_poc_DOC	dissolved organic carbon	$\mu\text{g L}^{-1}, \mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
eqpac_poc_DON	dissolved organic nitrogen	$\mu\text{g L}^{-1}, \mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
eqpac_poc_CTON	C to N ratio of particulate organic matter	ratio	ratio

† Note that not all cruises report all variables. Also, cruises 7,11 reported values in  $\mu\text{g L}^{-1}$ , while cruises 8,12 reported values in micromolar.

### 3.2.3 North Atlantic Bloom Experiment - NABE (Fig. 11c)

Nominal location:	47°N 20°W
Dates of Coverage:	NABE: intervals between April and August of 1989 BOFS: intervals between April and August of 1989 PRIME: Jun–Jul 1995; 6 week campaign in 1996
Data Sources:	(1) <a href="http://usjgofs.whoi.edu/jg/dir/jgofs/nabe/">http://usjgofs.whoi.edu/jg/dir/jgofs/nabe/</a> (2) BOFS Data Set on CD-ROM; (3) PRIME Data Set on CD-ROM; and also (4) Geoff Evans (from Fasham and Evans 1995.)

#### *Site-specific Methodology*

NABE cruise data are available from the US JGOFS site for: *Atlantis II* leg 4 (1989 Apr 20 – May 10); *Atlantis II* leg 5 (1989 May 15 – Jun 08); and *Endeavor* (1989 Jun 29 – July 6). Also used in this analysis are data obtained from the Plankton Reactivity in the Marine Environment (PRIME) Project Data Set (Hadziabdic and Cramer 1999), and from the Plankton Biogeochemical Ocean Flux Study (BOFS) (Lowry et al. 1994). Both of these data sets are available on CD-ROMS from the British Oceanographic Data Center (<http://www.bodc.ac.uk/>).

The method of calculating MLD at the NABE location is determined using a criterion of  $0.05 \sigma_\theta$  difference from the surface. This value is used because it produces similar MLDs to those determined by the BOFS method (Lowry et al. 1994) (Fig. 14). Lowry et al. (1994) determined MLD as the depth where a gradient of  $0.05^\circ\text{C m}^{-1}$  was sustained for at least 4 m. We calculate MLDs using the criteria of 0.02, 0.05, and  $0.125 \sigma_\theta$  (similarly to that used at other JGOFS sites). The  $0.05 \sigma_\theta$  criterion most closely matches that used for the BOFS calculations.

Bottle, primary production, and pigment data are provided for not only the US JGOFS cruises (NABE), but also for the U.K. cruises (BOFS, and PRIME). In addition, Geoff Evans' original data are provided, which are 0–30 m averages of  $\text{NO}_3$ ,  $\text{NH}_4$ , PON, zoo-plankton, and bacteria reported in terms of nitrogen ( $\text{mmol N m}^{-3}$ ); and chlorophyll in  $\text{mg m}^{-3}$ .

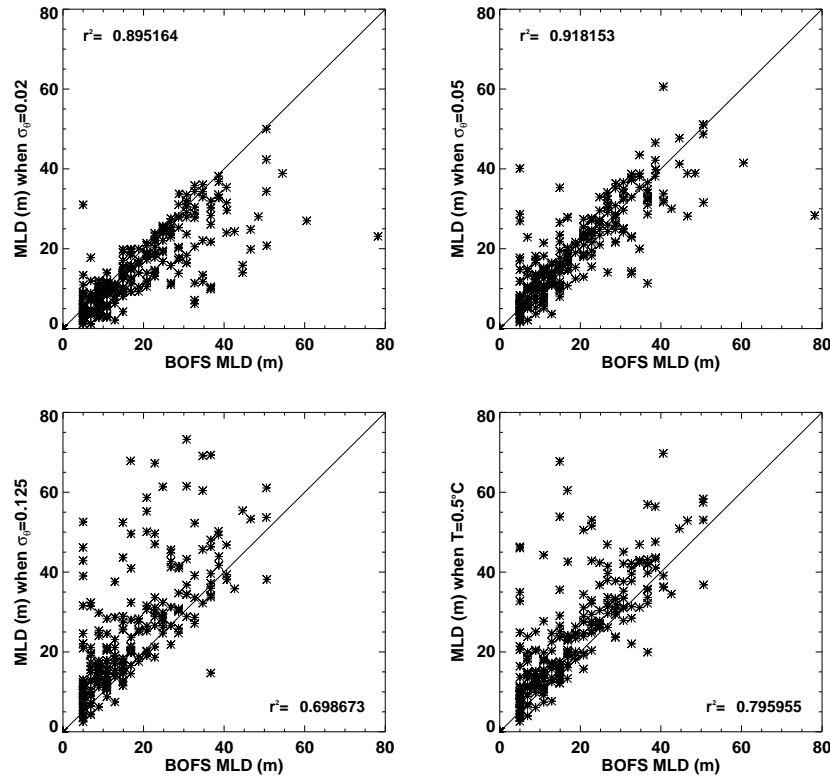


Fig. 14. Comparison of MLDs determined with the criteria:  $\sigma_\theta = 0.02, 0.05, 0.125$ ; and  $\text{Temp} = 0.5^\circ\text{C}$ , with those determined by BOFS.

## Description of reported variables

### NABE bottle data

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
nabe_b_TEMP	temperature	$^\circ\text{C}$	$^\circ\text{C}$
nabe_b_POTEMP	potential temperature	$^\circ\text{C}$	$^\circ\text{C}$
nabe_b_SAL	salinity	PSU	PSU
nabe_b_SIGMA_T	sigma-t		
nabe_b_O2	oxygen	$\text{ml L}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_NO3	nitrate	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_NO2	nitrite	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_PO4	phosphate	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_SIO3	silicate	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_NH4	ammonium	$\text{nmol L}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_POC	particulate organic carbon	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_PON	particulate organic nitrogen	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_TALK_b	total alkalinity (Brewer)	$\mu\text{eq kg}^{-1}$	$\mu\text{eq kg}^{-1}$
nabe_b_TALK_t	total alkalinity (Takahashi)	$\mu\text{eq kg}^{-1}$	$\mu\text{eq kg}^{-1}$
nabe_b_TCO2_b	total $\text{CO}_2$ (Brewer)	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_TCO2_t	total $\text{CO}_2$ (Takahashi)	$\mu\text{mol kg}^{-1}$	$\mu\text{mol kg}^{-1}$
nabe_b_PCO2_20	$\text{pCO}_2$ at $20^\circ\text{C}$ (Takahashi)	$\mu\text{atm}$	$\mu\text{atm}$

*NABE primary production data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
nabe_pp_PP	prim prod <i>Atlantis</i> 119,4	$\mu\text{molC L}^{-1} \text{inc\_time}^{-1}$	$\text{mmolC m}^{-3} \text{d}^{-1}$
	prim prod <i>Atlantis</i> 119,5	$\text{mgC m}^{-3} \text{d}^{-1}$	$\text{mmolC m}^{-3} \text{d}^{-1}$
	prim prod <i>Endeavor</i>	$\text{mgC m}^{-3} \text{d}^{-1}$	$\text{mmolC m}^{-3} \text{d}^{-1}$

*NABE bacteria data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
nabe_biol_BACT	bacterial abundance	$\text{cells} \times 10^9 \text{ L}^{-1}$	$\text{cells} \times 10^8 \text{ L}^{-1}$

*NABE pigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
nabe_pig_CHL_A	chlorophyll <i>a</i>	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$
nabe_pig_PERIDININ	peridinin	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$
nabe_pig_FUCOX_BUT	butanoyloxyfucoxanthin	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$
nabe_pig_FUCOX	fucoxanthin	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$
nabe_pig_FUCOX_HEX	hexanoyloxyfucoxanthin	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$
nabe_pig_DIADINOX	diadinoxanthin	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$
nabe_pig_DIATOX	diatoxanthin	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$
nabe_pig_ZEAX	zeaxanthin	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$
nabe_pig_CAROTENE	carotene	$\text{ng L}^{-1}$	$\text{mg m}^{-3}$

*Geoff Evans' Data †*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
nabe_e_NO3	$\text{NO}_3$	$\text{mmol N m}^{-3}$	$\text{mmol N m}^{-3}$
nabe_e_NH4	$\text{NH}_4$	$\text{mmol N m}^{-3}$	$\text{mmol N m}^{-3}$
nabe_e_PON	part. organic nitrogen	$\text{mmol N m}^{-3}$	$\text{mmol N m}^{-3}$
nabe_e_CHL	chlorophyll	$\text{mg m}^{-3}$	$\text{mg m}^{-3}$
nabe_e_BACT	bacteria	$\text{mmol N m}^{-3}$	$\text{mmol N m}^{-3}$
nabe_e_ZOO	zooplankton	$\text{mmol N m}^{-3}$	$\text{mmol N m}^{-3}$

† Data provided by Geoff Evans, are averages for the top 30 m

*BOFS bottle data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
bofs_b_TEMP	temperature	°C	degc
bofs_b_SAL	salinity	PSU	PSU
bofs_b_O2	dissolved O <sub>2</sub>	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
bofs_b_NO3	NO <sub>3</sub>	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
bofs_b_NO2	NO <sub>2</sub>	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
bofs_b_PO4	PO <sub>4</sub>	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
bofs_b_SI	SiO <sub>4</sub>	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
bofs_b_NH4	NH <sub>4</sub>	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
bofs_b_CPHYL	chlorophyll	$\text{mg m}^{-3}$	$\text{mg m}^{-3}$

*BOFS primary production data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
bofs_c14_PTOT	<sup>14</sup> C uptake primary production	$\text{mgC m}^{-2} \text{day}^{-1}\dagger$	$\text{mgC m}^{-2} \text{day}^{-1}\dagger$
bofs_c14_CHLTOT	total chlorophyll (sum: >5 $\mu\text{m}$ fraction >1–5 $\mu\text{m}$ fraction >0.2–1 $\mu\text{m}$ fraction in original BOFS C14DAT file)	$\text{mg m}^{-3}$	$\text{mg m}^{-3}$

<sup>†</sup> Note that BOFS provided primary production as the integrated value over the euphotic depth. Original units are retained here.

*BOFS pigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
bofs_pig_CHFL	fluorometric chl	mg m <sup>-3</sup>	mg m <sup>-3</sup>
bofs_pig_PHFL	fluorometric phaeopigments	mg m <sup>-3</sup>	mg m <sup>-3</sup>
bofs_pig_CHLSP	spectrophoto. chl	mg m <sup>-3</sup>	mg m <sup>-3</sup>
bofs_pig_PHSP	spectrophoto. phaeopigments	mg m <sup>-3</sup>	mg m <sup>-3</sup>
bofs_pig_CHLHPLC	HPLC chl <i>a</i>	μg L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_CHLC3	chlorophyll <i>c</i> <sub>3</sub>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_CHLC1C2	chlorophyll <i>c</i> <sub>1</sub> <i>c</i> <sub>2</sub>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_CHLB	chlorophyll <i>b</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_PERID	peridinin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_BUTAN	butanoyloxyfucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_FUCOX	fucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_HEXOXY	hexanoyloxyfucoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_DIADIN	diadinoxanthin	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_LUTEIN	lutein	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_BCAROT	β-carotene	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_CHLIDEA	chlorophyllide <i>a</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_PHORBA	phaeophorbide <i>a</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_PHORBL	phaeophorbide like	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_PHPHYTA	phaeophytin <i>a</i>	ng L <sup>-1</sup>	mg m <sup>-3</sup>
bofs_pig_PHPHYTL	phaeophytin like	ng L <sup>-1</sup>	mg m <sup>-3</sup>



*PRIME bottle data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
prime_b_TEMP	temperature (unspec. method)	°C	°C
prime_b_TEMP_CTD	temperature (CTD)	°C	°C
prime_b_TEMP_RT	temperature (rev. therm.)	°C	°C
prime_b_POT_TEMP	potential temperature	°C	°C
prime_b_SAL	salinity (unspec.)	PSU	PSU
prime_b_SAL_CTD	salinity (CTD)	PSU	PSU
prime_b_SAL_BENCH	salinity (bench salin.)	PSU	PSU
prime_b_O2_WINK	O <sub>2</sub> (Winkler)	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
prime_b_O2_BECK	O <sub>2</sub> (Beckmann)	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
prime_b_O2_SAT	O <sub>2</sub> saturation	%	%
prime_b_NO3_1	nitrate	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
prime_b_NO2_1	nitrite	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
prime_b_SIO3_1	silicate	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
prime_b_CHLA	chl <i>a</i> (unspec. method)	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
prime_b_CHLA_F0	chl <i>a</i> (in-situ fluorometry)	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
prime_b_CHLA_F1	chl <i>a</i> (GF/F filtered)	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
prime_b_CHLA_F2	chl <i>a</i> (0.4 $\mu\text{m}$ filtered)	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
prime_b_CHLA_F5	chl <i>a</i> (0.4 $\mu\text{m}$ filtered)	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
prime_b_PHAEO_F1	phaeopigments (GF/F filtered)	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
prime_b_DOC	dissolved organic C	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
prime_b_POC	particulate organic C	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
prime_b_PON	particulate total N	$\mu\text{mol L}^{-1}$	$\mu\text{mol kg}^{-1}$
prime_b_FE_TOTDISS	dissolved total iron	$\text{nmol L}^{-1}$	$\text{nmol kg}^{-1}$
prime_b_FE_LIG_TOTDISS	total dissolved iron complexed by ligands	$\text{nmol L}^{-1}$	$\text{nmol kg}^{-1}$
prime_b_FE_INORG DISS	total dissolved inorganic iron	$\text{nmol L}^{-1}$	$\text{nmol kg}^{-1}$
prime_b_MESOOOZOO_BIO	total mesozoopl. biomass	$\text{mg m}^{-3}$	$\text{mg m}^{-3}$
	computed from cell cnt		
prime_b_MICROZOOO_BIO	microzoopl. biomass	$\text{mgC m}^{-3}$	$\text{mgC m}^{-3}$
prime_b_MICROZOOO_CNT	microzoopl. abundance	$\text{cells ml}^{-1}$	$\text{cells ml}^{-1}$
prime_b_BACT_CNT	total bacteria	$\text{cells ml}^{-1}$	$\text{cells ml}^{-1}$

*PRIME primary production data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
prime_b_C_UPTAKE_INSITU	carbon uptake (in situ incubation)	$\text{mg m}^{-3} \text{d}^{-1}$	$\text{mmolC m}^{-3} \text{d}^{-1}$
prime_b_C_UPTAKE_INCUB	carbon uptake (nat. light incubation)	$\text{mg m}^{-3} \text{d}^{-1}$	$\text{mmolC m}^{-3} \text{d}^{-1}$
prime_b_CALC_MAX	calcification maximum	see note †	see note †
prime_b_PHOTO_MAX	photosynth. maximum	$\text{mgC mgChl}^{-1} \text{h}^{-1}$	$\text{mgC mgChl}^{-1} \text{h}^{-1}$

† both original and final units of maximum calcification are  $\text{mgC } (\mu\text{E m}^{-2} \text{s}^{-1})^{-1} \text{mgChl}^{-1} \text{h}^{-1}$ .

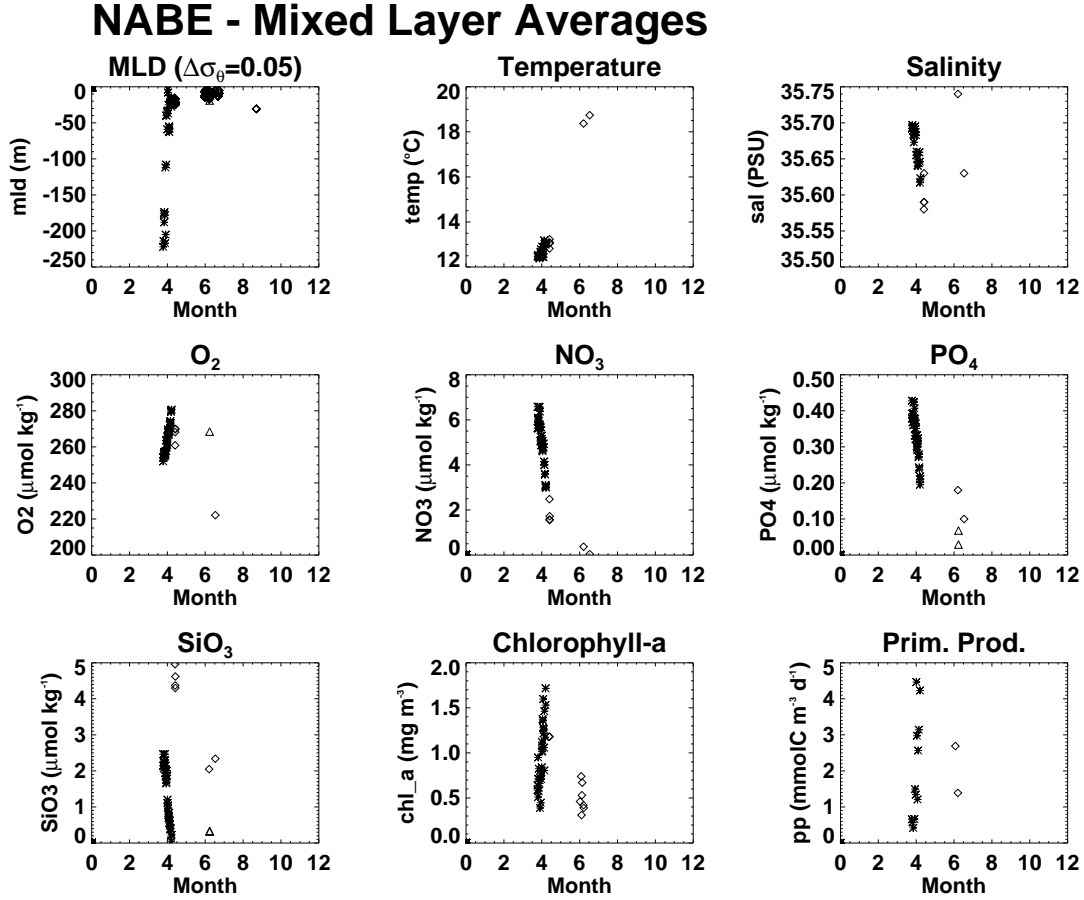


Fig. 15. Mixed layer averages for selected variables from North Atlantic Bloom Experiment, within a  $3^\circ$  grid cell centered at  $47^\circ\text{N}$ ,  $20^\circ\text{W}$ . Asterisk represents US JGOFS NABE samples; diamonds represent BOFS data; and triangles represent PRIME data.

### 3.2.4 and 3.2.5 SOUTHERN OCEAN REGIONS

US JGOFS sampled two major regions during the Antarctic Environment and Southern Ocean Process Study (AESOPS) cruises, the Antarctic Polar Front Zone (APFZ) and the Ross Sea. Cruises aboard the R/V *Roger A. Revelle*, which included two survey cruises and two process study cruises, were conducted almost entirely in the APFZ. Cruises aboard the R/V *Nathaniel B. Palmer* were conducted primarily in the Ross Sea area, although the site survey cruise (NBP-96-4) and the Benthic Process and Mooring Recovery cruise (NBP-98-2) were centered mainly in the APFZ.

Using data from all of the Palmer and Revelle cruises, we create separate data sets for the Ross Sea and the APFZ. Data are segregated according to geographic location: Those obtained from north of  $70^\circ\text{S}$  are included in the APFZ files, while those obtained from

south of 70°S are included in the Ross Sea files.

### 3.2.4 Antarctic Polar Front Zone (APFZ) (Fig. 11d)

Nominal Region:	60–70°S 165–175°W
Dates of Coverage:	Oct 1997 – Apr 1998
Data Source:	<a href="http://usjgofs.whoi.edu/jg/dir/jgofs/southern/">http://usjgofs.whoi.edu/jg/dir/jgofs/southern/</a> Directory KIWI includes AESOPS (KIWI) cruises near the APFZ (aboard the <i>Roger A. Revelle</i> ); for chart cruise tracks see: <a href="http://usjgofs.whoi.edu/images/aesops/revelle.gif">http://usjgofs.whoi.edu/images/aesops/revelle.gif</a>
Added Note:	a significant number of APFZ samples were also obtained on the <i>Revelle</i> cruises, see Ross Sea section for more information

### *Site-specific Methodology*

None.

### *Description of reported variables*

#### *APFZ bottle data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
apfz_b_TEMP	ctd temp	°C	°C
apfz_b_SAL_CTD	ctd salinity	PSS	PSS
apfz_b_SAL_BOT	bottle salinity (Autosal)	PSS	PSS
apfz_b_O2_1	oxygen (Winkler)	ml L <sup>-1</sup>	ml L <sup>-1</sup>
apfz_b_O2_2	oxygen (Winkler)	μmol kg <sup>-1</sup>	μmol kg <sup>-1</sup>
apfz_b_O2_3	oxygen (Winkler)	μmol L <sup>-1</sup>	μmol L <sup>-1</sup>
apfz_b_NO3	nitrate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
apfz_b_PO4	phosphate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
apfz_b_SiO4	silicate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
apfz_b_NO2	nitrite	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
apfz_b_NH4	ammonium	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>

*APFZ primary production data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
apfz_pp_CHL_A	chl <i>a</i> , fluorometric	mg m <sup>-3</sup>	mg m <sup>-3</sup>
apfz_pp_PP24	prim. prod., C assim.	mmolC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
apfz_pp_PB24	C assim. per unit chl <i>a</i>	mmolC mgChl <sup>-1</sup> d <sup>-1</sup>	mmolC mgChl <sup>-1</sup> d <sup>-1</sup>

*APFZ chlorophyll data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
apfz_pig_CHL_A	chl <i>a</i> – fluorometric	μg L <sup>-1</sup>	mg m <sup>-3</sup>

*APFZ HPLC pigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
apfz_hplc_ALLOX	alloxanthin	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CAROTENE_A	α-carotene	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CAROTENE_B	β-carotene	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CAROTENE_G	γ-carotene	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CHL_A1	chlorophyll <i>a</i> <sub>1</sub> ; [monovinyl chl <i>a</i> ]	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CHL_A1p	chlorophyll <i>a</i> ' <sub>1</sub>	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CHL_A_TOT†	chl <i>a</i> <sub>1</sub> , chl <i>a</i> ' <sub>1</sub> , chl <i>a</i> & allomerized chl <i>a</i> ' <sub>1</sub> ; [mono- vinyl chl <i>a</i> & chl <i>a</i> ']	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CHL_B1	chlorophyll <i>b</i> <sub>1</sub> ; [monovinyl chl <i>b</i> ]	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CHL_C12	chlorophyll <i>c</i> <sub>12</sub>	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CHL_C3	chlorophyll <i>c</i> <sub>3</sub>	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_P_CHLC3	phytolated chl <i>c</i> <sub>3</sub>	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_CHLIDE_A	chlorophyllide <i>a</i>	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_DIADINOX	diadinoxanthin	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_DIATOX	diatoxanthin	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_FUCOX	fucoxanthin	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_FUCOXANTHIOL	fucoxanthiol	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_FUCOX_BUT	19'-butanoyloxyfucoxanthin	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_FUCOX_HEX	19'-hexanoyloxyfucoxanthin	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_FUCOX_ISO1	fucoxanthin isomer 1	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_FUCOX_ISO2	fucoxanthin isomer 2	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_LUTEIN	lutein	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_PERIDININ	peridinin	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_VIOLAX	violaxanthin	μg L <sup>-1</sup>	mg m <sup>-3</sup>
apfz_hplc_ZEAX	zeaxanthin	μg L <sup>-1</sup>	mg m <sup>-3</sup>

† Note different reporting: Palmer cruises describe CHL\_A\_TOT as “Monovinyl chlorophyll *a* plus chlorophyllide *a*”; Revelle cruises describe it as “sum of chl\_a1, chlide\_a, chl\_a1' and allomerized chl a's”.

# APFZ bacteria data

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
apfz_bact_BACT_HET	heterotrophic bacteria abund.	cells $\times 10^9$ L $^{-1}$	cells $\times 10^8$ L $^{-1}$

# APFZ organic carbon and nitrogen data

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
apfz_toc_TOC	total organic carbon	$\mu\text{mol L}^{-1}$ , $\mu\text{mol kg}^{-1}$	$\mu\text{mol C kg}^{-1}$
apfz_doc_DOC	dissolved organic carbon	$\mu\text{mol L}^{-1}$	$\mu\text{mol C kg}^{-1}$
apfz_poc_POC	particulate organic carbon	$\mu\text{g L}^{-1}$	$\mu\text{mol C kg}^{-1}$
apfz_poc_PON	particulate organic nitrogen	$\mu\text{g L}^{-1}$	$\mu\text{mol N kg}^{-1}$
apfz_poc_cton†	carbon to nitrogen ratio of particulate organic matter	n/a	ratio

† C:N ratio is determined as the ratio of POC to PON.

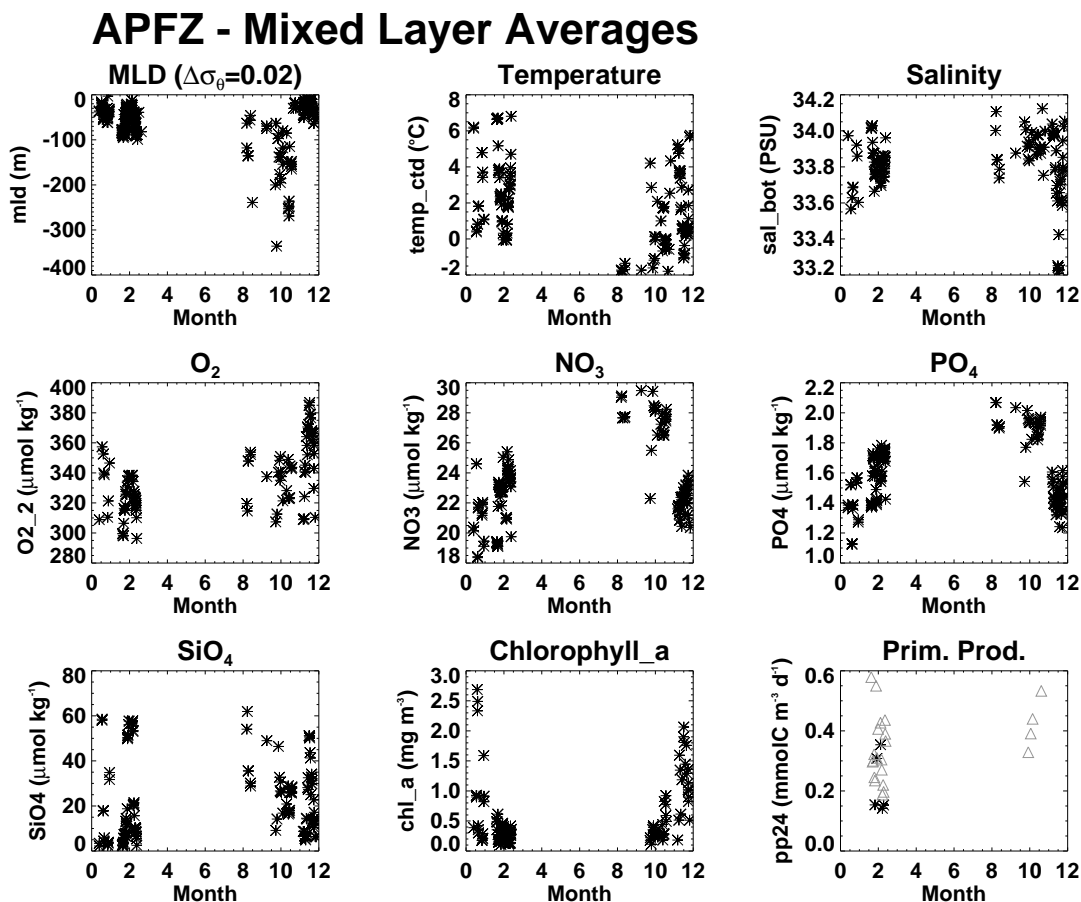


Fig. 16. Mixed layer averages for selected variables from Antarctic Polar Front Zone. For primary production, both in situ (asterisk) and on deck (triangle) incubations are shown.

### 3.2.5 Ross Sea (Fig. 11e)

Nominal Region:	70–76°S 168°E–175°W
Dates of Coverage:	Sep 1996 – Mar 1998
Data Source:	<a href="http://usjgofs.whoi.edu/jg/dir/jgofs/southern/">http://usjgofs.whoi.edu/jg/dir/jgofs/southern/</a> Directory ROSS includes data from the <i>Nathaniel B. Palmer</i> ); cruises; for chart of cruise tracks, see <a href="http://usjgofs.whoi.edu/images/aesops/palmer.gif">http://usjgofs.whoi.edu/images/aesops/palmer.gif</a>
Added Note:	Quite a few locations outside the Ross Sea were also sampled on the <i>Palmer</i> cruises (e.g., cruise nbp96_4, a site survey cruise, was primarily in the APFZ).

### *Site-specific Methodology*

None.

### *Description of reported variables*

#### *Ross Sea bottle data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
ross_b_TEMP	ctd temp	°C	°C
ross_b_SAL_CTD	ctd salinity	PSS	PSS
ross_b_SAL_BOT	bottle salinity (Autosal)	PSS	PSS
ross_b_O2_1	oxygen (Winkler)	ml L <sup>-1</sup>	ml L <sup>-1</sup>
ross_b_O2_2	oxygen (Winkler)	μmol kg <sup>-1</sup>	μmol kg <sup>-1</sup>
ross_b_O2_3	oxygen (Winkler)	μmol L <sup>-1</sup>	μmol L <sup>-1</sup>
ross_b_NO3	nitrate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
ross_b_PO4	phosphate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
ross_b_SiO4	silicate	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
ross_b_NO2	nitrite	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>
ross_b_NH4	ammonium	μmol L <sup>-1</sup>	μmol kg <sup>-1</sup>

*Ross Sea primary production data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
ross_pp_CHL_A	chl <i>a</i> , fluorometric	mgChl m <sup>-3</sup>	mgChl m <sup>-3</sup>
ross_pp_PP24	prim. prod., C assim	mmolC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
ross_pp_PB24	C assim per unit chl <i>a</i> on-deck incubation	mmolC mgChl <sup>-1</sup> d <sup>-1</sup>	mmolC mgChl <sup>-1</sup> d <sup>-1</sup>
ross_pp_od_CHL_A	chl <i>a</i> , fluorometric on-deck incubation	mgChl m <sup>-3</sup>	mgChl m <sup>-3</sup>
ross_pp_od_PP24	prim. prod., C assim on-deck incubation	mmolC m <sup>-3</sup> d <sup>-1</sup>	mmolC m <sup>-3</sup> d <sup>-1</sup>
ross_pp_od_PB24	C assim per unit chl <i>a</i> on-deck incubation	mmolC mgChl <sup>-1</sup> d <sup>-1</sup>	mmolC mgChl <sup>-1</sup> d <sup>-1</sup>

*Ross Sea bacteria data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
ross_bact_BACT_HET	heterotrophic bacteria abund.	cells×10 <sup>9</sup> L <sup>-1</sup>	cells×10 <sup>8</sup> L <sup>-1</sup>

*Ross Sea organic carbon and nitrogen data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
ross_toc_TOC	total organic carbon	μmol L <sup>-1</sup> , μmol kg <sup>-1</sup>	μmolC kg <sup>-1</sup>
ross_doc_DOC	dissolved organic carbon	μmol L <sup>-1</sup>	μmolC kg <sup>-1</sup>
ross_poc_POC	particulate organic carbon	μg L <sup>-1</sup>	μmolC kg <sup>-1</sup>
ross_poc_PON	particulate organic nitrogen	μg L <sup>-1</sup>	μmolN kg <sup>-1</sup>
ross_poc_cton†	carbon to nitrogen ratio of particulate organic matter	n/a	ratio

† C:N ratio is determined as the ratio of POC to PON.

*Ross Sea phaeopigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
ross_pig_CHL_A	chlorophyll <i>a</i>	μg L <sup>-1</sup>	mg m <sup>-3</sup>
ross_pig_PHAEO	phaeopigments	μg L <sup>-1</sup>	mg m <sup>-3</sup>
ross_pig_FO_TO_FA	ratio of fluoro. readings before (Fo) and after (Fa) acidification	ratio	ratio

*Ross Sea HPLC pigment data*

FILENAME	VARIABLE	ORIG UNITS	FINAL UNITS
ross_hplc_ALLOX	alloxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CAROTENE_A	$\alpha$ -carotene	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CAROTENE_B	$\beta$ -carotene	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CAROTENE_G	$\gamma$ -carotene	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CHL_A1	chlorophyll $a_1$ ; [monovinyl chl $a$ ]	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CHL_A1p	chlorophyll $a'_1$	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CHL_A_TOT†	chl $a_1$ , chl $a'_1$ , chlide $a$ & allomerized chl $a'$ ; [mono- vinyl chl $a$ & chlide $a$ ]	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CHL_B1	chlorophyll $b_1$ ; [monovinyl chl $b$ ]	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CHL_C	chl $c_1$ , chl $c_2$ & Mg 3,8 divinyl pheoporphyrin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CHL_C12	chlorophyll $c_{12}$	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CHL_C3	chlorophyll $c_3$	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_P_CHLC3	phytolated chl $c_3$	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_CHLIDE_A	chlorophyllide $a$	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_DIADINOX	diadinoxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_DIATOX	diatoxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_FUCOX	fucoxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_FUCOXANTHIOL	fucoxanthiol	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_FUCOX_BUT	19'-butanoyloxyfucoxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_FUCOX_HEX	19'-hexanoyloxyfucoxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_FUCOX_ISO1	fucoxanthin isomer 1	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_FUCOX_ISO2	fucoxanthin isomer 2	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_LUTEIN	lutein	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_PERIDININ	peridinin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_PRASINOX	prasinoxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_VIOLAX	violaxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$
ross_hplc_ZEAX	zeaxanthin	$\mu\text{g L}^{-1}$	$\text{mg m}^{-3}$

† Palmer and Kiwi cruises report CHL\_A\_TOT slightly differently: Palmer cruises describe it as “Monovinyl chlorophyll  $a$  plus chlorophyllide  $a$ ”; Revelle cruises describe it as “sum of chl- $a_1$ , chlide- $a$ , chl- $a_1'$  and allomerized chl  $a's$ ”.



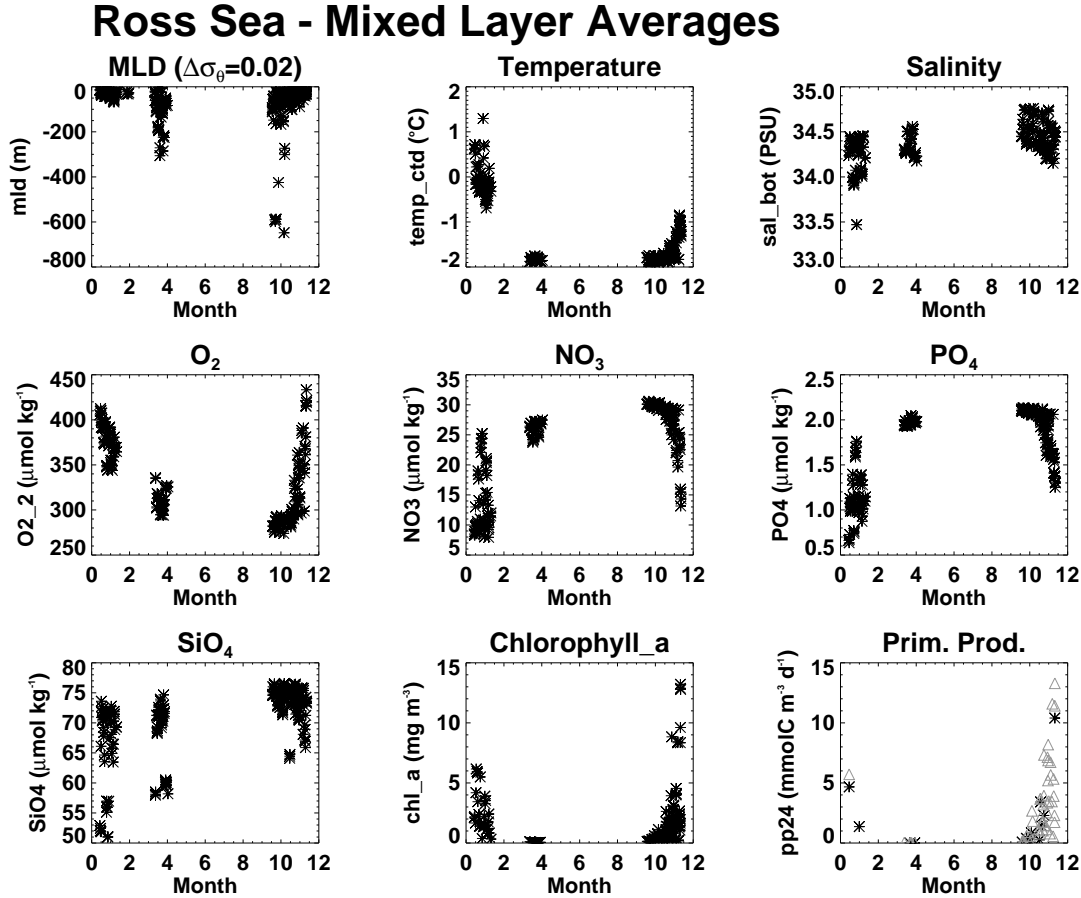


Fig. 17. Mixed layer averages for selected variables from the Ross Sea. For primary production, both in situ (asterisk) and on deck (triangle) incubations are shown.

## 4. GENERAL NOTES AND OBSERVATIONS

Probably the most important factor to take into account when using these data are the different methodologies applied at the various sites, as well as differences in how data are reported. This is addressed in detail earlier in this report for primary production measurements, but similar caution is recommended with other parameters as well.

For example, surface chlorophyll only is reported at Station P, while all other sites report chlorophyll concentrations for several depths. A comparison of the two measures illustrates that surface chlorophyll and mixed layer chlorophyll are highly correlated at all sites (Fig. 18); but this is certainly not true of all variables.

## Mixed Layer vs Surface Chlorophyll a

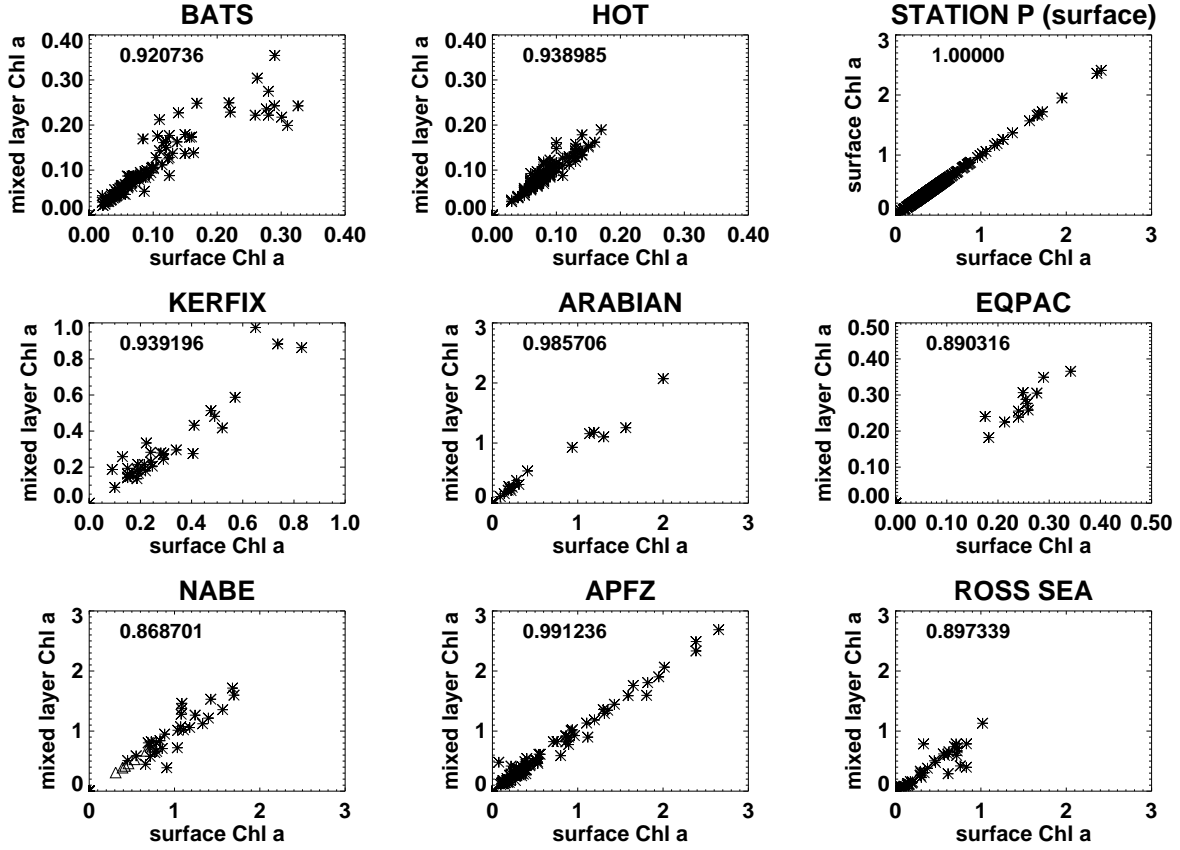


Fig. 18. Mixed layer chlorophyll concentration versus surface chlorophyll concentration ( $\text{mg m}^{-3}$ ). Regression coefficients are indicated for each site. Station P has surface values only, but is included for comparison with the other sites.

The ratio of nitrate and phosphate concentrations in the mixed layer varies considerably between sites (Fig. 19). However, caution should be exercised when interpreting  $\text{NO}_3:\text{PO}_4$  ratios, because nutrient concentrations often drop to very low levels (and at times below detection limits) at these sites.

A better way to examine the relationship between  $\text{NO}_3$  and  $\text{PO}_4$  is to use the  $\text{N}^*$  determination of Gruber and Sarmiento (1997):

$$\text{N}^* = \text{N} - 16\text{P} + \text{constant}$$

where:  $\text{constant} = 2.90 \mu\text{mol kg}^{-1}$  (Fig. 20). Among the time-series stations, BATS and HOT have a consistent surplus of  $\text{NO}_3$  relative to  $\text{PO}_4$ .  $\text{N}^*$  at KERFIX remains close to 0, but  $\text{N}^*$  concentrations at Station P are low. Low  $\text{N}^*$  concentrations generally indicate a net

## Nitrate and Phosphate

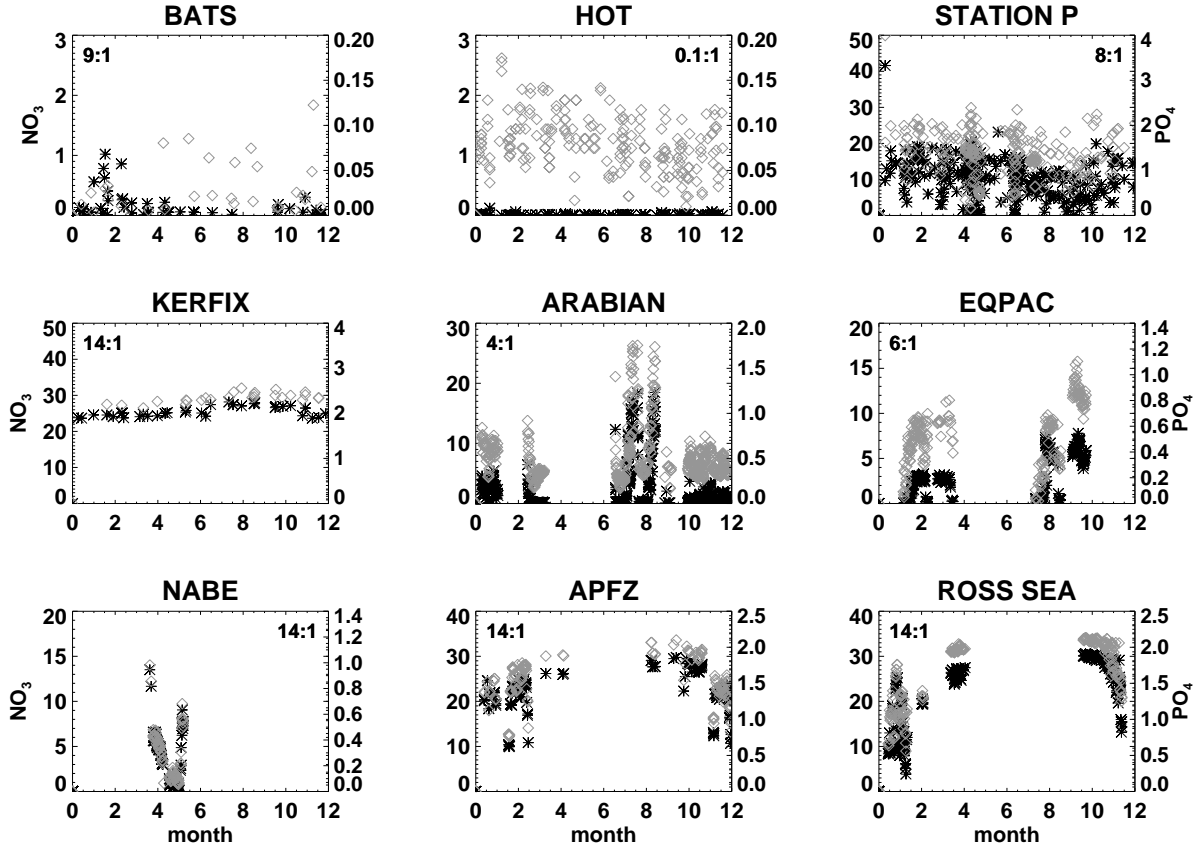


Fig. 19. Comparison of mixed layer averages of nitrate (asterisk) and phosphate (diamond) at the nine JGOFS sites. For ease of comparison, phosphate concentrations are plotted at the  $\text{NO}_3\text{:PO}_4$  Redfield ratio of 16x. The average  $\text{NO}_3\text{:PO}_4$  ratio is indicated for each site.

loss of  $\text{N}^*$  (e.g. due to denitrification), and high  $\text{N}^*$  is an indication of nitrogen fixation, or advection of fixed nitrogen from elsewhere. Whether denitrification is occurring at Station P, or whether the low  $\text{N}^*$  values represent a sampling problem is unclear. Most of the Process Study sites show a wide range of  $\text{N}^*$  values; this is partly due to the fact that the data shown were taken over a broader spatial area, and partly due to biological effects.

Also illustrated here is the between-site variation in (1) particulate organic carbon and dissolved organic carbon (Fig. 21); and (2) the C:N ratio of particulate organic matter (Fig. 22).

## Average $N^*$ in the MIXED LAYER

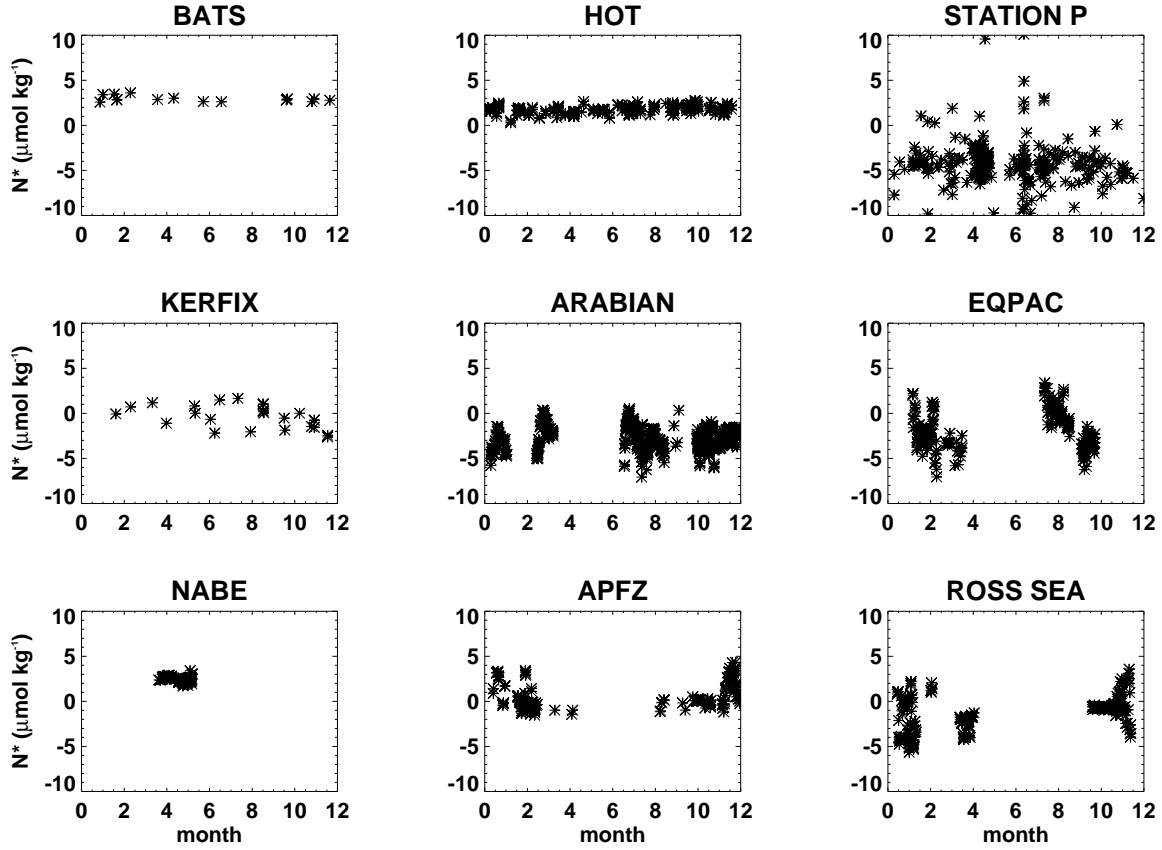


Fig. 20.  $N^*$  at nine JGOFS sites.  $N^*$  is calculated according to the equation of Gruber and Sarmiento (1997)  $N^* = N - 16P + 2.90 \mu\text{mol kg}^{-1}$ . Note that all data for the process study sites are included, which introduces spatial variation in  $N^*$  along cruise tracks.

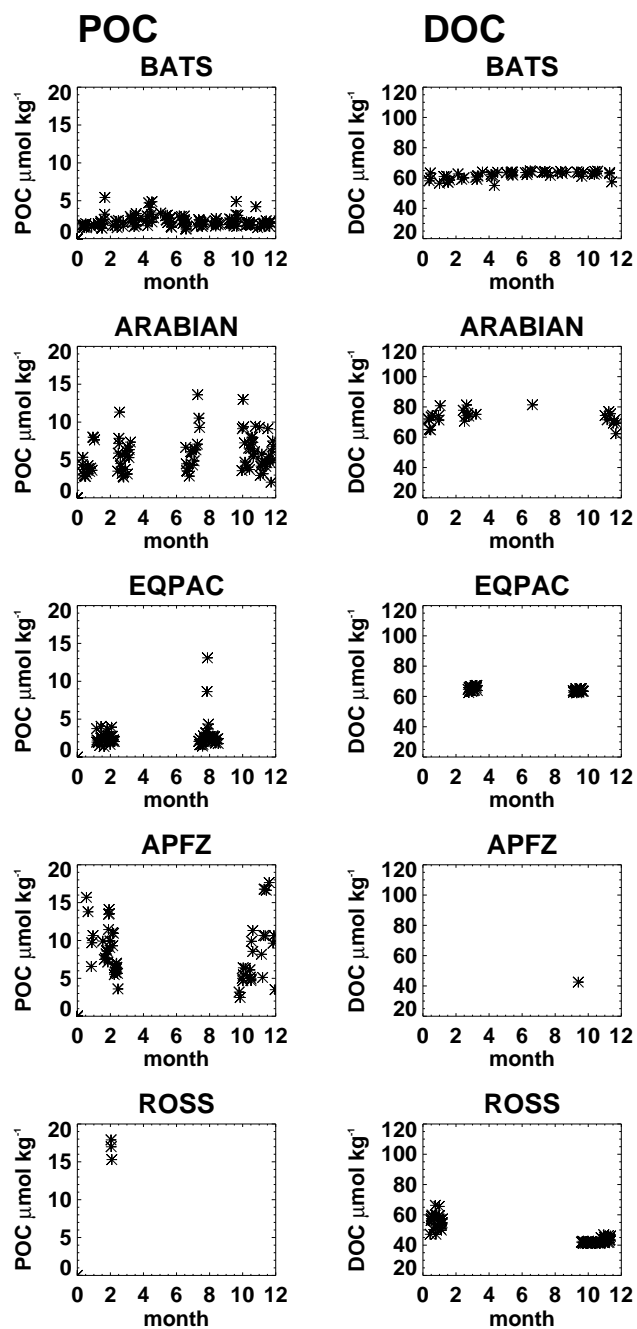


Fig. 21. Particulate organic carbon (POC) and dissolved organic carbon (DOC) at five sites. Note that DOC measurements for BATS and the Arabian Sea are not explicitly provided, and are calculated as difference between total organic carbon (TOC) and POC.

## C:N ratio of POM

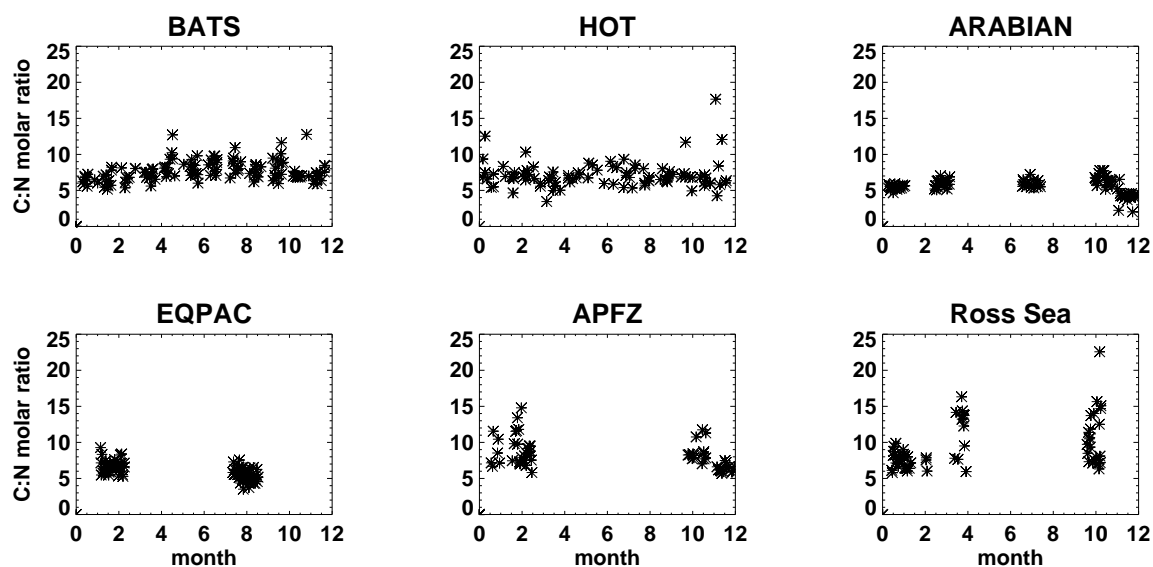


Fig. 22. C:N molar ratios of particulate organic matter at six sites. Note that the HOT ratio was determined for particulate matter (organic and inorganic components were not distinguished in the “particulate carbon” and “particulate nitrogen” data provided), while the ratio for other sites was determined for particulate organic matter.

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87 figs.

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## 6. OTHER REFERENCES AND RESOURCES

The following is a non-exhaustive list of resources, special issues, and other references that provide information about the time-series and process study sites included in this report. These are organized by site.

### GENERAL U.S. JGOFS AND INTERNATIONAL JGOFS

Knap, A., A. Michaels, A. Close, H. Ducklow, and A. Dickson (eds.), 1996. *Protocols for the Joint Global Ocean Flux Study (JGOFS) Core Measurements*. JGOFS Report No. 19, vi+170 pp. Reprint of the IOC Manuals and Guides No. 29, UNESCO 1994.

<http://usjgofs.who.edu:81/mzweb/contrib.htm>

Maintained by the U.S. JGOFS. Lists all U.S. JGOFS publications. Papers are listed alphabetically by first author.

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### HOT and BATS

Karl, D.M. and A.F. Michaels (eds.), 1996. Ocean Time Series: Results from the Hawaii and Bermuda Research Programs *Deep Sea Res. II* 43 (2–3).

Siegel, D., A.F. Michaels, and D. Karl (eds.), 2001. Interpretations of Open Ocean Biogeochemical Processes at the US JGOFS Bermuda and Hawaii Time Series Sites. *Deep Sea Res. II* 48 (8–9).

[http://www.bbsr.edu/users/ctd/data\\_reports.html](http://www.bbsr.edu/users/ctd/data_reports.html)

Maintained by the Bermuda Biological Research Station. Provides links to full text of BATS data reports and methods manuals.

<http://hahana.soest.hawaii.edu/hot/hotpub.html>

Maintained by School of Ocean and Earth Science and Technology, University of Hawaii. Lists (often with abstracts) HOT Publications, Data Reports and Manuals, Submitted Papers, Thesis and Dissertations, and Newsletters.

### STATION P

Boyd, P.W. and P.J. Harrison (eds.), 1999. Canadian JGOFS in the NE Subarctic Pacific. *Deep Sea Res. II* 46 (11–12).

*Contains numerous papers which address Station P.*

## **KERFIX**

Le Fèvre, J. and Tréguer, P. (eds.), 1998. Special Issue: Carbon Fluxes and Dynamic Processes in the Southern Ocean: Present and Past. *Journal of Marine Systems* 17 (1–4).

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## **ARABIAN SEA**

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## **EQUATORIAL PACIFIC**

Murray, J.W. (ed.), 1995. A U.S. JGOFS Process Study in the Equatorial Pacific *Deep*

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## **SOUTHERN OCEAN**

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guidance on primary production standards

Station P data

guidance on primary production standards

NABE data

KERFIX data

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**For:**

guidance on primary production standards

Station P data

Quality control

KERFIX mixed layer depths

KERFIX data

Station P data

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